

REVIEW

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# Ultra-high performance liquid chromatography-MS/MS (UHPLC-MS/MS) in practice: analysis of drugs and pharmaceutical formulations

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## Abstract

**Background:** UHPLC-MS/MS is connected in various research facilities for the qualitative and quantitative investigation of a pharmaceutical substance, pharmaceutical items, and biological specimen.

**Main body:** The commence review article is an endeavor to offer pervasive awareness around assorted aspects and details about the UHPLC-MS/MS and related techniques with the aim on practice to an estimation of medicinal active agents in the last 10 years. The article also focused on isolation, separation, and characterization of present impurity in drug and biological samples.

**Conclusion:** Review article compiles a general overview of medicinally important drugs and their analysis with UHPLC-MS/MS. It gives fundamental thought regarding applications of UHPLC-MS/MS for the study on safety limit. The summary of developed UHPLC-MS/MS methods gives a contribution to the future trend and limitations in this area of research.

**Key words:** UHPLC-MS/MS, UHPLC, Pharmaceutical analysis, Impurity, Bio-analytical

## Background

Ultra-high performance liquid-chromatography (UHPLC) covers liquid chromatography separations implementing columns enclose particles smaller than the 2.5–5  $\mu\text{m}$  sizes typically used in high-performance liquid chromatography (HPLC) [1]. UHPLC work on the same assumption as that of HPLC [2] and of which governing principle is that, as column packing particle size decrease, efficiency and thus resolution accretion [3]. Separations using column contain smaller particles display enhance efficiency per unit time [1, 2], but the efficiency cannot minimize at superior mobile phase flow rates or linear velocities [4]. After attribute, slighter particles, rapidity, and peak resolution can be absolute to new limits [5].

Since it is extremely well recognized from Van Deemter equations, the efficiency of the chromatographic process is proportional to particle size decrease [6]. Pursuant to his

model characterized band broadening, it clarifies by the connection between height equivalent of a theoretical plate (HETP) and linear velocity, is reliant on a diameter of particle packed into the analytical column [4, 6]. These accomplishments show the very considerable increase in resolution, sensitivity, and efficiency with the quicker outcome and a smaller amount expenditure of solvents which lowers the cost and make the technology environment friendly [7].

The hyphenated technique has turned out a precious method for the assessment of pharmaceuticals in various biological samples. The hyphenated technique is an attachment of the chromatographic system linked with the spectroscopic system with the right interface, such as LC-MS/MS [8]. It is well known that the detection profit appreciably from the performance characteristics of the UHPLC technology. As there is reduced in chromatography dispersion with an increased concentration of analyte will encourage improved source ionization efficiency [5].

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**Table 1** Acceptance criteria for UPLC-MS/MS

Characteristics	UHPLC
Particle size	< 2 $\mu\text{m}$
Maximum back pressure	1000 bars
Analytical column	UHPLC BEH $\text{C}_{18}$
Column dimensions	50 $\times$ 2.1 mm
Injection volume	2 $\mu\text{L}$
Column temperature	65 $^{\circ}\text{C}$
Total run time	1.5 min
USP resolution	3.4
Plate count	7500
Flow rate	0.6 mL/min

The hyphenated techniques usually applied for identification as well as quantification of the analyte when it is compared with other analytical methods [8]. At present era, employing of UHPLC gives the full benefits of chromatography applications for the separations using shorter columns, and superior flow rates for augmented rapidity, with greater resolution and sensitivity. The recognition criterion of UHPLC-MS/MS is shown in Table 1 [7]. The current review recapitulates the applications of UHPLC-MS/MS method for assessing drug in pharmaceutical as well as the biological matrix. In literature, there are several methods reported for analysis of

drugs using UHPLC-MS/MS, the major categorized such as antidiabetic, anticancer, antibiotics, cardiovascular, antiviral, nonsteroidal anti-inflammatory drug (NSAID), and others.

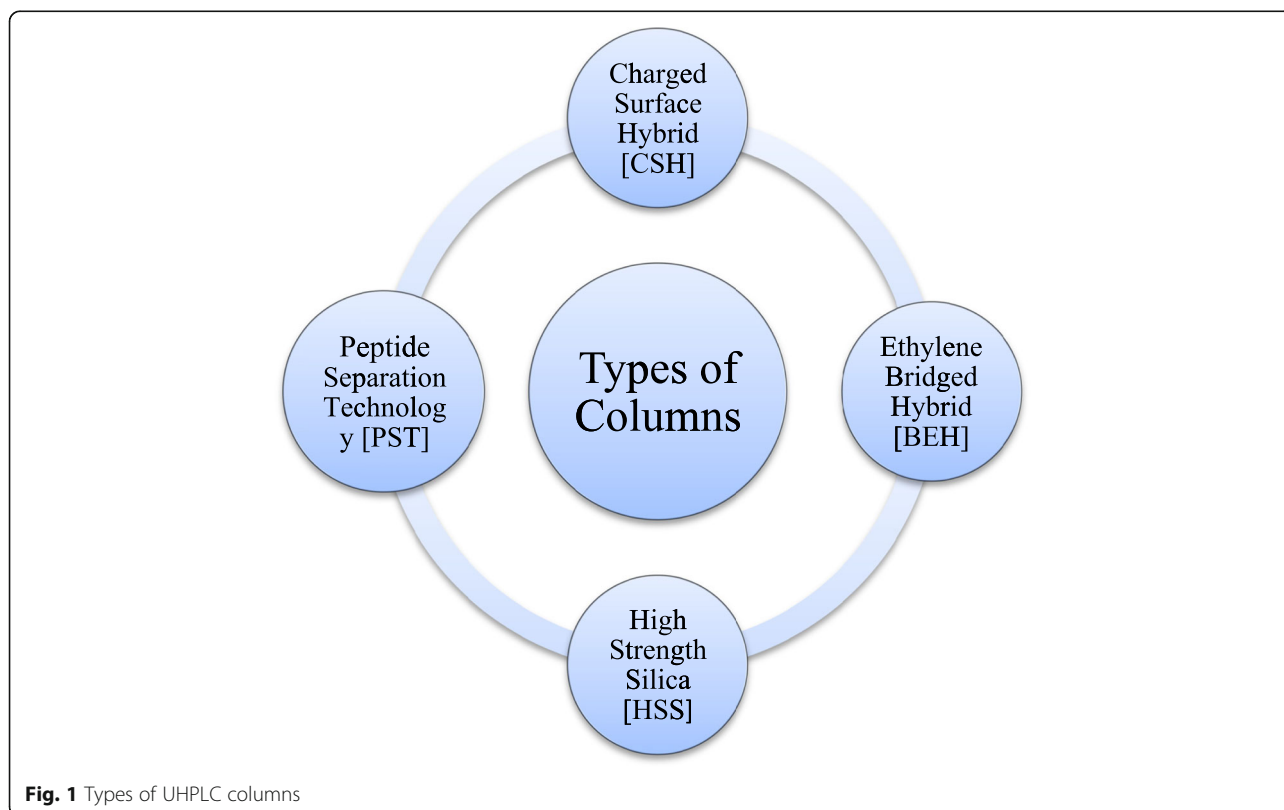
### Main text

#### UHPLC columns

The Acquity UHPLC column involved in the front line of liquid chromatography (LC) column development by giving higher quality chromatographic data's in less time. For use in applying up to 15000 psi (1000 bar), UHPLC columns are designed, certified, and tested [9]. However, different technologies produced distinct nature of columns actuality used in UHPLC is depicted in Fig. 1.

#### Charged surface hybrid

Third-generation particle technology was developed by waters, designed to recover sample loadability and peak tailing in uncertain ionic strength mobile phase composition. The charged surface hybrid retains the low level surface charge with 1.7  $\mu\text{m}$  particle size [9]. The basic charged surface hybrid (CSH)  $\text{C}_{18}$  column characteristics such as peak shape and increased loading capacity, mainly for basic compounds under low-pH, weak-ionic-strength mobile-phase [10]. The polyaromatic compound selective straight-chain-alkyl is especially used in Acquity UHPLC CSH Phenyl-Hexyl column and also it

**Fig. 1** Types of UHPLC columns

gives exceptional peak shape under different pH conditions [10]. The Acquity UHPLC CSH Fluoro-Phenyl columns show excellent selectivity for polar compounds, positional isomer, and halogenated compounds. This is due to a dipole-dipole, hydrogen-bonding, aromatic, and hydrophobic interaction [10].

#### Ethylene-bridged hybrid

The deficiency of mechanical strength or efficacy is essential to complete the potential speed, sensitivity, and resolution capabilities for primary generation methyl-hybrid particle of xTerra columns. Hence, there is need of a column with a new pressure-tolerant particle needed to create. An innovative, additional hybrid material columns were set that contains an ethylene-bridged hybrid material. It displayed enhanced efficiency, pH range, and strength as compared to first-generation columns [11]. The developed ethylene-bridged hybrid (BEH) columns fixed polar group attached to the silyl functionality with a C<sub>6</sub> alkyl and also for UHPLC BEH phenyl columns [7].

#### High strength silica

High strength silica (HSS) is another type of column used in UHPLC. In UHPLC, high pore volume UHPLC particles do not acquire the mechanical stability necessary to hold up the high pressure innate of UHPLC separations [9]. For that, there is established a novel silica particle and appropriate morphology required to give long and lifetime efficiency UHPLC column at high pressure likely 1000 bars. HSS particle technology is the modern automation; 1.8 μm UHPLC HSS particles are designed and exclusively for separations using UHPLC [12]. To overcome trouble during separation and retention of small water-soluble and polar organic molecules during reversed phase separation, Acquity UHPLC HSS T<sub>3</sub> columns were developed. The Acquity UHPLC HSS C<sub>18</sub> selectivity for bases (SB) columns is a non-encapped, low-coverage silica-based C<sub>18</sub> chemistry that alternate selectivity for water-soluble compounds influenced by silanophilic interactions. The enhanced silanol activity of the HSS C<sub>18</sub> SB column result in greater retention of basic compounds; due to secondary interactions with residual silanols while simultaneously

**Table 2** Present perspective on UHPLC-MS/MS methods

Author(s)	Title	Year	Ref
Tatsuya Higashi, Kazutake Shimada, Toshimasa Toyo'oka	Advances in determination of vitamin D related compounds in biological samples using liquid chromatography–mass spectrometry: a review	2009	[20]
Aurelie Roux, Dominique Lison, Christophe Junot, Jean-François Heilier	Applications of liquid chromatography coupled to mass spectrometry-based metabolomics in clinical chemistry and toxicology: a review	2010	[21]
Alexander A. Zoerner, Frank-Mathias Gutzki, Sandor Batkai, Marcus May, Christin Rakers, Stefan Engeli, Jens Jordan, Dimitrios Tsikas	Quantification of endocannabinoids in biological systems by chromatography and mass spectrometry: a comprehensive review from an analytical and biological perspective	2011	[22]
Fabio Gosetti, Eleonora Mazzucco, Maria Carla Gennaro, Emilio Marengo	Ultra high performance liquid chromatography tandem mass spectrometry determination and profiling of prohibited steroids in human biological matrices. a review	2012	[23]
Marcus Bantscheff & Simone Lemeer & Mikhail M. Savitski & Bernhard Kuster	Quantitative mass spectrometry in proteomics: critical review update from 2007 to the present	2012	[24]
Mark R. Viant, Ulf Sommer	Mass spectrometry based environmental metabolomics: a primer and review	2012	[25]
Maria-Jose Motilva, Aida Serra, Alba Macia	Analysis of food polyphenols by ultra high-performance liquid chromatography coupled to mass spectrometry: an overview	2013	[26]
Nikalje Anna Pratima, Baheti Shraddha, Sayyad Zibran	Review of ultra performance liquid chromatography and its applications	2013	[27]
Maria Joao Gouveia, Paul J. Brindley, Lucio Lara Santos, Jose Manuel Correia da Costa, Paula Gomes, Nuno Vale	Mass spectrometry techniques in the survey of steroid metabolites as potential disease biomarkers: a review	2013	[28]
Muchena J. Kailemia, L. Renee Ruhaak, Carlito B. Lebrilla, and I. Jonathan Amster	Oligosaccharide analysis by mass spectrometry: a review of recent developments	2013	[29]
Linda Kortz, Juliane Dorow, Uta Ceglarek	Liquid chromatography–tandem mass spectrometry for the analysis of eicosanoids and related lipids in human biological matrices: a review	2014	[30]

**Table 3** Bioanalytical/metabolite studies

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/ metabolite studies or pharmaceutical	Mode of analysis	Ref
<b>Antibiotics agents</b>							
Amoxicillin, Ampicillin, cefuroxime, cefazolin, ceftriaxime, meropenem, piperacillin, clavulanic acid, and tazobactam	BEH C <sub>18</sub>	Water (A) Acetonitrile (B) both containing 0.1% formic acid 5% solution (B) at 0.5 min to 55% at 4.0 min and column was rinsed with 95% solution (B) for 0.5 min and re-equilibrated to starting conditions for 1 min	Triple-quadrupole	PP	Biological	Gradient	[33]
Cefuroxime lysine	BEH C <sub>18</sub>	Acetonitrile: 0.1% formic acid in 10 mM Ammonium Acetate (40:60, v/v).	Quadrupole	LLE	Biological	Isocratic	[34]
Clarithromycin	BEH C <sub>18</sub>	(A) 50 mM ammonium Acetate solution and (B) Acetonitrile. 0.0 min 50 (A):50(B), 1.5 min 10(A): 90(B), 1.7 min 10(A): 90(B), 1.8 min 50(A):50(B), 2.0 min 50(A):50(B)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[35]
Indolinone derivative	HSS T <sub>3</sub> column	Water (A) and (B) acetonitrile; both containing 0.1% formic acid 0–0.5 min, (B) 2%; 0.5–2 min, (B) 2–100%; 2–2.5 min, (B) 100%; 2.5–2.6 min, (B) 100–2%; 2.6–4 min, (B) 2%	Mass detector	PP	Biological	Gradient	[36]
Ertapenem	Agilent Zorbax Eclipse Plus C <sub>18</sub>	Water : Acetonitrile (95:5% v/v) both containing 0.1 % formic Acid	Triple-quadrupole	LLE	Biological	Gradient	[37]
Amikacin, flucloxacillin, meropenem, penicillin G and vancomycin	RP <sub>18</sub> column	Water (A) and methanol (B), both containing 4% of formic acid 0–0.3 min, 20% (B); 0.3–1.0 min, 50% (B); 1.0–2.0 min 50% (B); 2.0–2.7 min, 75% (B), 2.7–5.3 min, 75% (B) 5.3–5.5 min, 20% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[38]
Cephalomannine	BEH C <sub>18</sub>	0.1% formic acid in water (A) and acetonitrile (B) 0–0.6 min, 20–85% (B); 0.6–1.8 min, 85% (B); 1.8–2.0 min, 80–30% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[39]
Tedizolid and linezolid	BEH C <sub>18</sub>	(A) 0.1% formic acid in water and (B) Acetonitrile 0–0.3 min, 20–95% (B); 0.3–1.5 min, 95–95% (B); 1.5–1.6 min, 95–20% (B); 1.6–3.0 min, 20–20% (B)	Mass spectrometer	PP	Biological	Gradient	[40]
<b>Anticancer agents</b>							
Amrubicin	BEH C <sub>18</sub>	(A) Acetonitrile: water, (1:99% v/v), 5 mM formic acid and (B) acetonitrile: water, (99:1% v/v), also containing 5 mM formic acid 0.00 min, 100(A) :0(B); 0.30 min, 0(A):100 (B); 0.70 min, 0(A):100 (B); 1.50 min, 100(A): 0(B)	Triple-quadrupole	SPE	Metabolite	Gradient	[41]

**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/ metabolite studies or pharmaceutical	Mode of analysis	Ref
Solasonine	HSS T <sub>3</sub> column	(A) 0.1% formic acid and (B) acetonitrile. 0–0.5 min, 15% (B); 0.5–3.5 min, 15–75% (B); 3.5–4.0 min, 75–85% (B); 4.0–5.0 min, 85% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[42]
Berberrubine	BEH C <sub>18</sub>	(A) 0.1% formic acid and (B) acetonitrile. 1.5 min, 30–60% (B); within 0.5 min, 95% (B); then decreased to 30% (B) within in 0.1 min, then maintained 30% for 0.4 min	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[43]
Pomalidomide	BEH C <sub>18</sub>	Acetonitrile: 10 mM Ammonium Acetate (80:20% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[44]
Docetaxel	BEH C <sub>18</sub>	(A) Water containing 0.5% formic acid (pH 2.23) and (B) 0.5% formic acid in acetonitrile (pH 2.24). 0–0.9 min, 90% (A); 0.9–1.6 min, 100% (B); 1.6–1.7 min, 90% (A), 1.7–2.5 min, 90% (A)	Quadrupole	LLE	Biological	Gradient	[45]
Indolinone derivative	HSS T <sub>3</sub> column	Water (A) and (B) acetonitrile, both containing 0.1% formic acid. 0–0.5 min, 2% (B); 0.5–2 min, 2–100% (B); 2–2.5 min, 100% (B); 2.5–2.6 min, 100–2% (B); 2.6–4 min, 2% (B)	Mass detector	PP	Biological	Gradient	[36]
Lenalidomide	BEH C <sub>18</sub>	Acetonitrile: water: formic acid (65:35: 0.1% v/v/v)	Triple-quadrupole mass spectrometer	PP	Biological	Isocratic	[46]
Paclitaxel	BEH C <sub>18</sub>	Methanol–0.1% aqueous formic acid (75: 25% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[47]
Irinotecan and 7-ethyl-10 hydroxy camptothecin (SN-38)	BEH C <sub>18</sub>	(A) Ammonium acetate Buffer (10 mM; pH 3.5) with 0.1% formic acid and (B) acetonitrile. 0 min, 20% (B); 3 min, 20–70% (B); 0.2 min, 70–98% (B), 0.7 min, 98% (B); 0.2 min, 98–20% (B)	Quadrupole	SPE	Biological	Gradient	[48]
Anticancer c-Met inhibitor LS-177	C <sub>18</sub> column	(A) Acetonitrile and (B) 0.1% formic acid in water. 0 min, 20% (A); 1 min, 80% (A); 2 min, 80% (A); 2.5 min, 20% (A); 3 min, 20% (A)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[49]
TM-2	C <sub>18</sub> column	(A) Acetonitrile and (B) 2 mmol/L ammonium acetate in water. 1.5 min, 60–90% (A); 2.8 min, 90% (A); 3.5 min, 90–60% (A)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[50]
Methotrexate polyglutamates	BEH C <sub>18</sub>	(A) 10 mM Ammonium bicarbonate buffer. With 5 mM of the desired ion-pair adjusted to pH 7.5 with formic acid and (B) acetonitrile. 1 min, 90% (A); 9 min, 70% (A); 30% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[51]

**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/metabolite studies or pharmaceutical	Mode of analysis	Ref
Ciprofloxacin, cyclophosphamid, tamoxifen, and azathioprine	HSS T <sub>3</sub>	(A) 0.1 % formic acid (A) and (B) acetonitrile: 0 min, 90% (A) and 5% (B); 10 min, 0% (A) and 100 % (B)	Triple-quadrupole mass spectrometer	SPE	Pharmaceutical	Gradient	[52]
Temozolomide	BEH C <sub>18</sub>	(A) Water (containing 0.1% formic acid and 10 mM ammonium acetate) and (B) acetonitrile: 0 min, 20% (B); 2 min, 20–50% (B); 1 min, 50% (B); 1 min, 50–20% (B)	Quadrupole mass spectrometer	PP	Biological	Gradient	[53]
Dendrobine	BEH C <sub>18</sub>	(A) 0.1% formic acid and (B) acetonitrile: 0–1.0 min, 20–40 (B); 2 min, 40–95% (B), 0.5 min, 95% (B); then decreased to 20% (B) within 0.1 min, maintained at 20% (B) for 0.4 min	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[54]
Simotininb	Zorbax SB-C <sub>8</sub> column	0.1 % formic Acid with 10 mM ammonium formate: methanol (20: 80% v/v)	Mass detector	LLE	Biological	Isocratic	[55]
Nintedanib	BEH C <sub>18</sub>	(A) acetonitrile and (B) 0.1% formic acid in water	Triple-quadrupole mass spectrometer	PP	Metabolite	Gradient	[56]
Letrozole	C <sub>18</sub> column	0–0.3 min, 20–95% (A); 0.3–2.0 min, 95–95% (A); 2.0–2.1 min, 95–20% (A); 2.1–3.0 min, 20–20% (A) (A) Acetonitrile and 0.1% formic acid in water (B); 0 min, 20% (A); 80% (B); 0.3 min, 20% (A); 80% (B); 2 min, 95% (A); 5% (B); 2.5 min, 95% (A); 5% (B); 2.6 min, 20 % (A); 80% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[57]
Parthenolide	BEH C <sub>18</sub>	(A) Acetonitrile and 0.1% formic acid in water (B); 0–0.3 min, 20–95% (B); 0.3–1.5 min, 95–95% (B), 1.5–1.6 min, 95–20% (B)	Triple-quadrupole mass spectrometer	SPE	Biological	Gradient	[58]
Tamoxifen	BEH C <sub>18</sub>	(A) 10 mM Ammonium formate in water and acetonitrile (B) both are containing 0.1% formic acid 0.0 min, 70%(A): 30%(B); 9.0 min, 48% (A): 52%(B); 9.01 min, 48% (A): 52%(B); 9.50 min, 70% (A): 30%(B); 13 min 70% (A): 30%(B)	Triple-quadrupole mass spectrometer	PP	Metabolite	Gradient	[59]
Vincristine	HSS T <sub>3</sub> column	(A) Methanol and (B) water containing 0.2% formic acid. 3 min, 30–95% (A); and returned to initial condition within 1 min	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[60]
Palonosetron	HSS T <sub>3</sub> column	Methanol : water containing 0.1% formic acid 80: 20% v/v	Triple-quadrupole	LLE	Biological	Isocratic	[61]

**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/ metabolite studies or pharmaceutical	Mode of analysis	Ref
Kurarinone	BEH C <sub>18</sub>	(A) Acetonitrile and 0.1% formic acid in water (B) 0–0.3 min, 30–30% (A); 0.3–1.0 min, 30–90% (A); 1.0–2.0 min, 90–90% (A); 2.0–2.1 min, 90–30% (A); 2.1–3.0 min, 30–30% (A)	Triple-quadrupole	LLE	Biological	Gradient	[62]
Antiviral agents							
Elbasvir	BEH C <sub>18</sub>	(A) Acetonitrile and water (containing 5.0 mM ammonium acetate with 0.01% acetic acid, pH 4.5) (B) 0.0–0.5 min, 90–90% (A) 0.5–1.2 min, 90–5% (A); 1.2–2.0 min, 5–5% (A); 2.0–2.2 min, 5–90% (A); 2.2–3.0 min, 90–90% (A)	Triple-quadrupole	PP	Biological	Gradient	[63]
Dactasvir	HSS C <sub>18</sub>	10 mM Ammonium formate, pH 3.5 and acetonitrile (50:50% v/v)	Quadrupole	LLE	Biological	Isocratic	[64]
Ribavirin, sofosbuvir	BEH C <sub>18</sub>	(A) Acetonitrile and 0.1% formic acid in water (B) 0–0.5 min, 10–10% (A); 0.5–1.0 min, 10–95% (A); 1.0–2.0 min, 95–95% (A); 2.0–2.1 min, 95–10% (A); 2.1–3.0 min, 10–10% (A)	Triple-quadrupole mass spectrometer	PP	Metabolite	Gradient	[65]
Atazanavir, Darunavir and Ritonavir	C <sub>18</sub> column	(A) 10 mM Ammonium Formate, pH 4.0, adjusted with formic acid in water, and acetonitrile (B). 0.0–0.8 min, 50%(A) and 50%(B); 0.8–1.2 min, 30%(A) and 70%(B)	Mass detector	PP	Biological	Gradient	[66]
Oleanolic acid	C <sub>18</sub> column	10 mM Ammonium Acetate with 0.1% formic acid in water: acetonitrile (10: 90% v/v)	Triple-quadrupole mass spectrometer	SPE	Biological	Isocratic	[67]
Antifungal agent							
Tizoxanide	BEH C <sub>18</sub>	Aq. ammonium formate buffer 10 mM, adjusted to pH 3.0 with formic acid, and acetonitrile (40: 60% v/v)	Mass detector	LLE	Biological	Isocratic	[68]
Voriconazole	BEH C <sub>18</sub>	Acetonitrile and water containing 1% formic acid (45: 55% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[69]
Amphotericin B	BEH Shield RP <sub>18</sub> column	(A) Methanol: acetonitrile (50 : 50% v/v) containing 0.1% formic acid; and (B) 10 mM ammonium formate (pH 3 ± 0.2), containing 0.2% formic acid and 1% acetonitrile 0.0–2.0 min, 35–90% (A); 2.0–2.7 min, 90–35% (A)	Triple-quadrupole mass spectrometer	SPE	Biological	Gradient	[70]

**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/metabolite studies or pharmaceutical	Mode of analysis	Ref
Fluconazole, Itraconazole, Hydroxyitraconazole, Posaconazole, Voriconazole	C <sub>18</sub> column	(A) 10 mM Ammonium formate in ultrapure water with 0.1% formic acid and acetonitrile with 0.1% formic acid (B) (A) 10 mM Ammonium formate in water containing 0.1% formic acid, (B) 1% acetic acid in methanol, and (C) 1% acetic Acid in acetonitrile. 0–0.9 min, 90% (A) and 10% (B); 0.9–1.9 min, 5% (A) and 95% (B); 1.9–2 min, 100% (B); 2–3.6 min, 100% (C); 3.6–3.7 min, 90% (A) and 10% (B); 3.7–5.0 min, 90% (A) and 10% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Isocratic	[71]
Posaconazole	BEH C <sub>18</sub>	(A) 10 mM Ammonium formate in water containing 0.1% formic acid, (B) 1% acetic acid in methanol, and (C) 1% acetic Acid in acetonitrile. 0–0.9 min, 90% (A) and 10% (B); 0.9–1.9 min, 5% (A) and 95% (B); 1.9–2 min, 100% (B); 2–3.6 min, 100% (C); 3.6–3.7 min, 90% (A) and 10% (B); 3.7–5.0 min, 90% (A) and 10% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[72]
Cardiovascular agents Atorvastatin, Ezetimibe	BEH C <sub>18</sub>	(A) 0.1% Formic acid in water : (B) acetonitrile (95: 5% v/v) 0.0–0.4 min, 95% (A) and 5%(B); 0.4–1.1 min 10 %(A) and 90% (B); 1.1 min, 95% (A) and 5% (B)	Triple-quadrupole	LLE	Biological	Gradient	[73]
4-hydroxyphenyl carvedilol	BEH C <sub>18</sub>	Acetonitrile- 4.0 mM ammonium formate, pH 3.0 adjusted with 0.1% formic acid (78: 22% v/v)	Triple-quadrupole mass spectrometer	SPE	Metabolite	Isocratic	[74]
Ivabradine, Reboxetine, Metoprolol	BEH C <sub>18</sub>	(A) Water and (B) methanol, both containing 2 mM ammonium acetate.0.0–0.5 min, 20% (B); 0.5–3.0 min, 20–90% (B); 3.0–3.5 min, 90% (B) 3.5–4.5 min, 20% (B)	Quadrupole mass spectrometer	LLE	Biological	Gradient	[75]
Meldonium	BEH HILIC column	Water: acetonitrile: formic acid (adjusted to pH 3.0 with 12.5% ammonia hydroxide) (25: 70: 5% v/v/v)	Triple-quadrupole mass spectrometer	PP	Biological	Isocratic	[76]
Candesartan, Hydrochlorothiazide	C <sub>18</sub> Phenomenex	Methanol : Acetonitrile 1 (15: 85% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[77]
Doxazosine	BEH C <sub>18</sub>	(A) and (B) consisted of 0.05 (w/v) pentadecafluorooctanoic acid in acetonitrile and 0.05 (w/v) pentadecafluorooctanoic acid in water. 1.0–1.45 min, 10–99% (A); 1.45–1.55 min, 99–10% (A); 1.55–2.0 min, 10% (A)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[78]
Lercanidipine	BEH C <sub>18</sub>	70% Acetonitrile in water containing 0.2% v/v formic acid	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[79]
Aliskiren Hemifumarate, Amlodipine Besylate and Hydrochlorothiazide	BEH C <sub>18</sub>	0.1% Formic acid in ammonium acetate buffer (0.02 M, pH 3.5) and Methanol (25: 75% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[80]



**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/metabolite studies or pharmaceutical	Mode of analysis	Ref
Irbesartan	BEH C <sub>18</sub>	Acetonitrile: Methanol: 10 mM Ammonium acetate (70:15: 15% v/v/v)	Triple-quadrupole	SPE	Biological	Isocratic	[81]
Amlodipine, benazepril and benazeprilat	BEH C <sub>18</sub>	(A) 0.1% Formic acid and Acetonitrile (B) 0–0.2 min, 95% (A) and 5% (B); 0.2–0.3 min, 0%(A) and 100%(B); 0.3–2.0 min, 0%(A) and 100%(B); 2.0–2.5 min, 95% (A) and 5% (B), 2.5 min, 95% (A) and 5% (B)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[82]
Candesartan	Betasil C <sub>8</sub> column	Methanol : ammonium tri-floro acetate buffer with formic acid (60: 40% v/v)	Mass detector	PP	Biological	Isocratic	[83]
N-butylscopolamine	C <sub>18</sub> column	Acetonitrile: 5 mM ammonium acetate: 0.1% formic acid (90: 10% v/v)	Triple-quadrupole mass spectrometer	SPE	Biological	Isocratic	[84]
Udenafil	BEH C <sub>18</sub>	Acetonitrile: 0.1% formic acid (75: 25% v/v)	Quadrupole mass spectrometer	LLE	Metabolite	Isocratic	[85]
Alprostadil	BEH C <sub>18</sub>	(A) Acetonitrile and water (B), containing 0.1% formic acid 0.0–0.6 min, 30% (A); 0.6–0.8 min, 60% (A); 1.4 min, 30% (A)	Quadrupole mass Spectrometer	LLE	Biological	Gradient	[86]
Analgesic agents							
Sumatriptan	Hypersil gold C <sub>18</sub>	Methanol:water (90:10% v/v)	Mass detector	PP	Biological	Isocratic	[87]
Sufentanil	BEH C <sub>18</sub>	Acetonitrile : water (4:55% v/v)	Triple-quadrupole mass spectrometer	SPE	Biological	Isocratic	[88]
Mitragynine	BEH HILIC column	10 mM Ammonium formate buffer containing 0.1% formic acid: acetonitrile (15:85% v/v)	Quadrupole	LLE	Biological	Isocratic	[89]
Tapentadol and tapentadol-O-glucuronide	BEH C <sub>18</sub>	0.01 M Ammonium formate (adjusted to pH 4 using formic acid) (A) and methanol (B). For Tapentadol-0.00 min, 80% (A):20%(B); 0.85 min, 65%(A):35%(B); 0.90 min, 2%(A):98% (B); 1.25 min, 2%(A):98% (B); 1.30 min, 80% (A): 20%(B); 1.60 min, 80% (A): 20%(B) Tapentadol-O-glucuronide-0.0 min, 80% (A): 20%(B); 0.50 min, 70% (A): 30%(B); 0.55 min, 2% (A): 98%(B); 1.05 min, 2% (A): 98%(B); 1.10 min, 80% (A): 20%(B); 1.50 min, 80% (A): 20%(B)	Triple-quadrupole mass spectrometer	SPE	Biological	Gradient	[90]
Dexmedetomidine	BEH C <sub>18</sub>	0.1% Formic acid in water (A) and 0.1% formic acid in acetonitrile (B): 0 min, 25%(B); 3.0 min, 25%(B); 3.1 min, 95%(B); 5.0 min, 95%(B); 8.0 min, 2%(B)	Triple-quadrupole mass spectrometer	SPE	Biological	Gradient	[91]

**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/metabolite studies or pharmaceutical	Mode of analysis	Ref
Nalbuphine	BEH HILIC column	Acetonitrile : water (83:17% v/v) that contained 0.2% formic acid and 4 mM ammonium formate	Triple quadrupole mass spectrometer	LLE	Biological	Isocratic	[92]
Buprenorphine and norbuprenorphine	BEH C <sub>18</sub>	Ammonium formate buffer 20 mM adjusted to pH 3.05 with formic acid (A) and acetonitrile (B) (90:10% v/v) 0.0 min, 10 % (B); 2.5 min, 25% (B); 3.2 min, 25% (B); 5.0 min, 30% (B)	Quadrupole mass spectrometer	PP	Biological	Gradient	[93]
Anti-steroid							
Testosterone and 5-dihydrotestosterone	BEH C <sub>18</sub>	(A) Water, (B) consisted of 2 mM ammonium acetate (native pH) in acetonitrile: water (98:2% v/v) 0.1 min, 0% (B); 1.0 min, 30% (B); 2.0 min, 32% (B); 3.4 min, 32% (B)	Triple-quadrupole	LLE	Biological	Gradient	[94]
Triamcinolone acetonide palmitate and triamcinolone acetonide	Phenomenex Luna C <sub>18</sub> column	(A) Acetonitrile and (B) 0.1% ammonia solution. 0–1.3 min, 34% (A); 1.3–1.5 min, 34–95% (A); 1.5–5.0 min, 95% (A); 5.0–5.2 min, 95–34% (A); 5.2–5.5 min, 34% (A)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[95]
Anti-inflammatory (NSAID)							
Hesperetin	Chiralpak IA-3	(A) Water with 0.1% CHOOH and (B) acetonitrile with 0.1% CHOOH. 0 min, 33% (B); 0.5 min 33% (B); 5 min 40% (B); 5.1 min, 100% (B); 6 min, 100% (B); 6.1 min 33% (B); 8 min 33% (B)	Triple-quadrupole	SPE	Biological	Gradient	[96]
Epiafzelechin a flavan-3-ol with osteoprotective	BEH C <sub>18</sub>	Water (A) and acetonitrile (B) both containing (0.1% Formic Acid v/v) 1–11 min, 0–70% (B); 0–1 min, 100 % (A); 11–16 min, 100% (A)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[97]
E6005, Phosphodiesterase 4 inhibitor	BEH C <sub>18</sub>	(A) Water, acetonitrile: 1 mol/L ammonium formate (95:5:0: 5% v/v/v) and (B) water: acetonitrile: 1 mol/L ammonium formate (100:500: 5% v/v/v). Gradient-2.0 min, 5–95% (B); Isocratic- 2.0 min, 95% (B); 1.5 min, 5% (B)	Triple-quadrupole mass spectrometer	SPE	Metabolite	Gradient/ Isocratic	[98]
Naringin	BEH C <sub>18</sub>	Acetonitrile-0.4% acetic acid (80: 20% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[99]
Ibuprofen	BEH C <sub>18</sub>	Water: methanol (35:65% v/v) both containing 10 mM ammonium acetate and 0.1% formic acid. 0–12 min, 35% (A) and 65% (B); 12.01–14 min 100% (B); 15.1 min, 35% (A) and 65% (B)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[100]

**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/ metabolite studies or pharmaceutical	Mode of analysis	Ref
Avicularin	BEH C <sub>18</sub>	0.1% Formic acid in water (A) and (B) acetonitrile. 0–0.50 min, 30–80% (B); 0.50–1.40 min, 80 % (B); 1.40–1.60 min, 80–30% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[101]
Hupehnine	BEH C <sub>18</sub>	0.1% Formic acid (A) and Acetonitrile (B) 0.0–2.0 min, 40–90 % (B); maintained for 0.5 min, 90%(B); 90–40% (B) within 0.1 min and then maintained 0.4 min, for 40% (B)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[102]
Picfeltaeransins IA and IB	C <sub>18</sub> column	Methanol:water (70 : 30% v/v)	Mass detector	PP	Biological	Isocratic	[103]
Diclofenac sodium	BEH C <sub>18</sub>	Acetonitrile–0.1% ammonium hydroxide aqueous solution (20:80% v/v). 0.0 min, 20% (A); 0.6–1.5 min, 50% (A); 1.6–2.0 min, 20% (A)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[104]
21-hydroxy deflazacort	BEH C <sub>18</sub>	Acetonitrile and 4.0 mM ammonium formate, pH 3.5 (90:10% v/v)	Triple-quadrupole mass spectrometer	SPE	Biological	Isocratic	[105]
Acetaminophen and dihydrocodeine	BEH C <sub>18</sub>	(A) 0.1% formic acid in water and (B) Acetonitrile 0–0.3 min, 50–95% (B); 0.3–0.9 min, 95–95% (B); 0.9–1.0 min, 95–50% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[106]
Anti-diabetes							
Glipizide	BEH C <sub>18</sub>	(A) Acetonitrile and water containing 1% formic acid (B). 0.0 min, 30 % (A); 70%(B); 0.3 min, 75% (A);25% (B); 0.7 min, 75% (A) 25% (B); 1.0 min, 30% (A); 70% (B)	Triple-quadrupole mass spectrometer	SPE	Biological	Gradient	[107]
Lysophosphatidylcholine	BEH C <sub>18</sub>	(A) 0.1% Formic acid in water, (B) 0.1% formic acid in acetonitrile, 0 min, 100% (A); 37 min, 100% (B)	Mass detector	PP	Metabolite	Gradient	[108]
Miglitol	C <sub>18</sub> inertsil column	Acetonitrile and 2 mM Ammonium Acetate (pH 3.5, 80: 20% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[109]
Exenatide	C <sub>18</sub> column	0.2% Formic acid (A) and methanol (B) 1.5 min, 10% (B); 1.0 min, 10–90% (B); 0.5 min, 90% (B) a linear decrease from 90–10% (B) in 0.1 min	Mass detector	SPE	Biological	Gradient	[110]
Anti-depressant							
Vortioxetine	BEH C <sub>18</sub>	0.1% Formic acid in water (A) and acetonitrile (B). 0–0.5 min, 20–95% (A); 0.5–1.9 min, 95–95% (A); 1.9–2.0 min, 95–20% (A); 2.0–3.0 min, 20–	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[111]

**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/metabolite studies or pharmaceutical	Mode of analysis	Ref
Bupropion	BEH Phenyl column	20% (A) (A) Ammonium formate buffer (2 mM, pH 4.0) and (B) acetonitrile. 3.5 min, 95% (A); 4.2 min, 70% (A); 4.7 min, 68.5% (A); 4.8 min, 5% (A) 5.7 min, 95% (A)	Triple-quadrupole mass spectrometer	SPE	Metabolite	Gradient	[112]
Venlafaxine and O-desmethylvenlafaxine	BEH C <sub>18</sub>	10 mmol/L Ammonium acetate and methanol (85: 15% v/v)	Triple-quadrupole	SPE	Biological	Isocratic	[113]
Paroxetine	Hypersil Gold C <sub>18</sub> column	Acetonitrile and 0.1% Glacial acetic acid (50: 50% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[114]
Antiepileptic agents Carbamazepine, carbamazepine-10, 11-epoxide, gabapentin, lamotrigine, levetiracetam	BEH Phenyl column	5 mM Ammonium bicarbonate buffer pH 7.9 (A) and methanol (B) 0.00 min, 97.5% (A); 2.5%(B); 1.00 min, 95.0% (A); 5.0% (B); 2.00 min, 0.0% (A); 30.0% (B); 3.60 min, 50.0% (A); 50.0% (B); 4.00 min, 50.0% (A); 50.0% (B); 4.30 min, 20.0% (A); 80.0% (B); 4.75 min, 20.0% (A) 80.0% (B); 4.85 min, 97.5% (A); 2.5% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[115]
Haloperidol, olanzapine, quetiapine, fluoxetine, paroxetine, bromazepam, lorazepam, risperidone and benzodiazepines	HSS T <sub>3</sub>	Acetonitrile (A) and 0.1% formic acid in water (B) 0–1 min, 75% (B); 1–6 min, 75–40%(B); 6 min, 75% (B)	Triple-quadrupole mass spectrometer	SPE	Biological	Gradient	[116]
Oxcarbazepine	BEH C <sub>18</sub>	Acetonitrile–10 mM ammonium acetate (85:15% v/v)	Triple-quadrupole mass spectrometer	SPE	Biological	Isocratic	[117]
Benzodiazepine 21	BEH C <sub>18</sub>	0.1% Formic acid in water (A) and 0.1% formic acid in Methanol (B). 70% A–30% B, ramping to 65% B over 2.5 min, 70% B over 0.75 min, 77% B over 0.75 min, then to 95% B over 0.05 min and holding at 95% B for 1 min, returning to 30% B and re-equilibrating for 2 min	Mass detector	LLE	Biological	Gradient	[118]
Aripiprazole	Agilent Eclipse Plus C <sub>18</sub>	(A) 0.1% Formic acid aqueous and (B) acetonitrile. 0–1.2 min, 10–40% (B); 1.2–1.6 min, 40–100% (B); 1.6–1.9 min, 100% (B); 1.91–2.0 min, 10% (B)	Mass detector	SPE	Biological	Gradient	[119]
CNS Levodopa, Carbidopa	BEH C <sub>18</sub>	(A) 5 mM Ammonium formate/formic acid at 1000/1 (v/v) and (B)	Triple-quadrupole and electro chemical	PP	Biological	Gradient	[120]

**Table 3** Bioanalytical/metabolite studies (Continued)

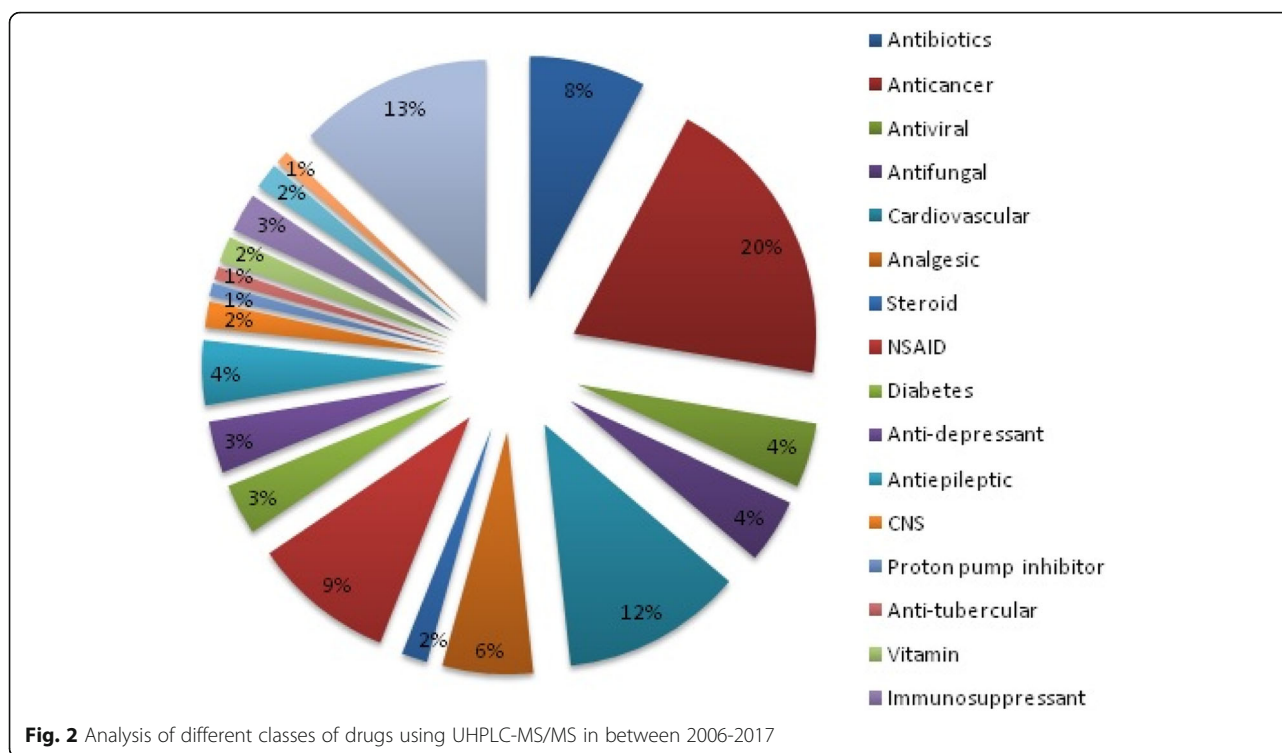
UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/metabolite studies or pharmaceutical	Mode of analysis	Ref
Acetylcholine, histamine (HA)	BEH HILIC column	acetone: 0.2 min, 25% (B); 0.2–1.8 min, 25–50% (B); 90% (B) for 0.9 min (A) 0.2% Formic acid and 20 mM ammonium formate in Water (B) acetonitrile, 0 min, 90% (B); 0–1.25 min, 10–70% (B)	Triple-quadrupole mass spectrometer	PP	Metabolite	Gradient	[121]
Proton pump inhibitor Esomeprazole, Rabeprazole and Levosulpiride	Hypersil gold C <sub>18</sub>	(A) 2 mM Ammonium formate in water and (B) acetonitrile, 0.01 min, 10% (B); 0.20 min, 10% (B); 0.50 min, 90% (B); 0.70 min, 90% (B); 0.80 min, 10% (B); 1.00 min, 10% (B)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[122]
Anti-tubercular Ethambutol, isoniazid and rifampicin	HSS T <sub>3</sub>	Water: methanol (50: 50% v/v)	Quadrupole	PP	Biological	Isocratic	[123]
Vitamin Thiamin, riboflavin, Flavin, adenine dinucleotide, nicotinamide, and pyridoxal	HSS T <sub>3</sub>	10 mM Ammonium formate (A) and acetonitrile (B), 0 min, 95% (A); 1 min, 65% (A); 2 min, 45% (A); 2.1–3 min, 5% (A), 3.1–4 min, 95% (A)	Mass detector	LLE	Biological	Gradient	[124]
Uracil and dihydrouracil	HSS T <sub>3</sub> column	(A) 0.1% (v/v) Formic acid in UPLC-grade water and (B) 0.1% (v/v) formic acid in UPLC-grade acetonitrile, 0–0 min, 0% (B); 3.0–3.2 min, 0–90% (B); 3.2–3.7 min, 90% (B); 3.7–5 min, 0% (B)	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[125]
Immunosuppressant Tacrolimus	BEH C <sub>18</sub>	10 mM Ammonium acetate, pH 6.00, adjusted with Formic acid and methanol (5:95% v/v)	Triple quadrupole mass spectrometer	SPE	Biological	Isocratic	[126]
Typhaneoside and Isoflammetin-3-O-Neohesperidoside	BEH C <sub>18</sub>	(A) Acetonitrile and water (containing 0.1% formic acid) (B) 0 min, 80% (A); 1.7 min, 68% (A); 3.2 min 30% (A); 4.0 min, 20% (A)	Mass detector	Pollen typhae extract	Biological	Gradient	[127]
Mycophenolic acid	BEH C <sub>18</sub>	Acetonitrile and 10 mM Ammonium formate, pH 3.0 (75:25% v/v)	Mass detector	PP	Biological	Isocratic	[128]
Anticoagulant Warfarin	BEH C <sub>18</sub>	(A) 0.2% Formic acid and 1% acetonitrile in water; and (B) 0.2% formic acid in acetonitrile, 80–70% (A) in 1.5 min and then changed linearly to 35% A for another 1.5 min, and remained	Mass detector	PP	Biological	Gradient	[129]

**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/metabolite studies or pharmaceutical	Mode of analysis	Ref
Clopidogrel	BEH C <sub>18</sub>	for a further 0.8 min before changing back to 80% (A) Acetonitrile and water containing 0.15% formic acid (75: 25% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[130]
Expectorant							
Hupehenine	BEH C <sub>18</sub>	(A) 0.1% Formic acid and acetonitrile (B), 0–2.0 min, 40–90% (B); maintained for 90% (B) for 0.5 min, then decreased to 40% within 0.1 min, then maintained at 40% for 0.4 min	Triple-quadrupole mass spectrometer	LLE	Biological	Gradient	[102]
Multicomponent analysis of drug							
Gestodene, Ethinyl estradiol	BEH C <sub>18</sub>	Water (A) and acetonitrile (B) (both containing 0.1% formic acid), 1.5 min, 70% (B); 70–98% (B) within 1.1 min and returned to 70% (B)	Quadrupole	LLE	Biological	Gradient	[131]
Liensinine, Isoliensinine, and Neferine	BEH C <sub>18</sub>	(A) Water with 0.1 formic acid and acetonitrile (B), 0–1.0 min, 15–90% (B); 1.0–1.8 min, 90% (B); 1.8–2.0 min, 90–15% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[132]
Pinaverium Bromide	BEH C <sub>18</sub>	Acetonitrile: 5 mM ammonium formate (80:20% v/v)	Triple-quadrupole mass spectrometer	PP	Biological	Isocratic	[133]
Pimpinellin, Isopimpinellin and Phellopterin	BEH C <sub>18</sub>	Methanol: ammonium acetate (65:35% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[134]
Endocannabinoids 2-arachidonoyl glycerol (2AG), 1-arachidonoyl glycerol (1AG), and anandamide	BEH C <sub>18</sub>	(A) Water contained 2 mM ammonium acetate and (B) methanol and contained 2 mM ammonium acetate: 0.0–0.5 min, 75% (B); 0.5–5.0 min, 79% (B); 5.0–5.5 min, 90% (B); 5.5–6.5 min, 75% (B)	Tandem mass spectrometry	PP	Biological	Gradient	[135]
Paenoniflorin, albiflorin, ferulic acid, tetrahydropalmatine, protopine, typhaneoside, senkyunolide I	BEH C <sub>18</sub>	(A) 0.1% Formic acid and (B) acetonitrile, 0–1 min, 5–10% (B); 1–6 min, 10–30% (B); 6–7 min, 30–40% (B); 7–8 min, 40–95% (B); 8–10 min, 95–5% (B)	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[136]
Caffeine, Tolbutamide, Metoprolol, Dapsone	BEH HILIC C <sub>18</sub>	Acetonitrile: water containing 0.1% formic acid (15:85% v/v)	Triple-quadrupole mass spectrometer	LLE	Biological	Isocratic	[137]
Xylazine, Free Morphine, Codeine, 6 Acetylmorphine, Cocaine, Benzoyllecgonine in Postmortem	HSS T <sub>3</sub>	0.1% Formic acid in water (A) and acetonitrile (B), 0 min, 90% (A); 10% (B); 0.48 min, 90% (A); 10% (B); 0.50 min, 100% (A); 0% (B); 0.60 min, 85% (A); 15% (B); 2.0 min, 66% (A); 34% (B); 2.10 min, 5%	Triple-quadrupole mass spectrometer	SPE	Biological	Gradient	[138]

**Table 3** Bioanalytical/metabolite studies (Continued)

UPLC (drug)	Column	Mobile phase system	Detector	Extraction method	Biological or/metabolite studies or pharmaceutical	Mode of analysis	Ref
Bupropion, metoprolol, midazolam, phenacetin, omeprazole and tolbutamide	BEH C <sub>18</sub>	(A): 95% (B); 2.25 min, 90% (A); 10% (B); 2.50 min, 90% (A); 10% (B) Acetonitrile (A) and water (B) (containing 0.1% formic acid) 0.3–1.8 min, 30–60% (A); increasing to 95% over 0.2–0.5 min, then decreased to 30% within 0.1 min, and maintained at 30% for 0.4 min.	Triple-quadrupole mass spectrometer	PP	Biological	Gradient	[139]
Ketoconazole and Voriconazole on the Pharmacokinetics of Oxcarbazepine	BEH C <sub>18</sub>	Acetonitrile (A) and water (B) (containing 0.1% formic acid) 0–1 min, 20–80% (A); 1–1.5 min, 80–80% (A), 1.5–2 min 80–95% (A); 2–2.5 min, 95–95% (A); 2.5–2.6 min, 95–20% (A)	Triple-quadrupole mass Spectrometer	PP	Metabolite	Gradient	[140]
Neochlorogenic acid, chlorogenic acid, cryptochlorogenic acid, and geniposide	BEH C <sub>18</sub>	0.2% Acetic acid and methanol. 0–4.50 min, 15–30% (B); 4.50–5.50 min, 30–65% (B); 5.50–6.00 min, 65–15% (B); 6.00–7.50 min, 15–15% (B)	Triple-quadrupole mass Spectrometer	PP	Biological	Gradient	[141]
Evodiamine and Rutacarpine	BEH C <sub>18</sub>	Acetonitrile (A) and water (B) both containing 0.2% Formic acid. 0 min, 40% (A); 60% (B); 0–0.6 min, 70% (A); 30% (B); 0.6–1.0 min, 70% (A); 30% (B); then returned to 40% (A)/60% (B)	Triple-quadrupole mass Spectrometer	LLE	Biological	Gradient	[142]
Bosentan and glimepiride	BEH C <sub>18</sub>	(A) 0.1% Formic acid in water and (B) acetonitrile. 0–0.5 min, 35% (B); 0.5–1.0 min, 35–80% (B); 1.0–1.5 min, 80–35% (B)	Triple-quadrupole	PP	Biological	Gradient	[143]
Amoxicillin and prednisolone	BEH C <sub>18</sub>	0.1% Formic acid in water (A) and 0.1% formic acid in acetonitrile (B). 0–1 min, 98% (A); 1–1.5 min, 98–15% (A); 1.5–2.5 min, 15% (A); 2.5–3.0 min, 15–2% (A); 3.0–3.5 min, 2–98% (A); 3.5–5.5 min, 98% (A)	Triple-quadrupole mass Spectrometer	SPE	Biological	Gradient	[144]
Fenofibric acid and berberine	Eclipse XDB C <sub>18</sub> column	(A) Acetonitrile and (B) water (containing 0.1% formic acid). 0.0 min, 30 % (A); 70% (B); 0.3 min, 30% (A); 70% (B); 0.5 min, 50% (A); 50% (B); 1.0 min, 90% (A); 10% (B); 2.5 min, 90% (A); 10% (B); 2.6 min, 30% (A); 70% (B); 3.0 min, 30% (A); 70% (B)	Triple-quadrupole mass Spectrometer	LLE	Biological	Gradient	[145]

**Table 4** Stability-indicating method UHPLC-MS/MS

Drug	Acid degradation	Base degradation	Photo-degradation	Oxidative degra.	Detection	Ref
Conditions	0.1 to 1 N	0.1 to 2 M	Light	H <sub>2</sub> O <sub>2</sub> and KMnO <sub>4</sub>	–	–
Fluticasone Propionate	7 days	(50 °C) 24 h	–	–	Thioester, Carboxylic ester, alcohol, ketone, diol	[147]
Duloxetine	(50 °C) 24 h	7 days	–	–	Ether,1-naphthol	
Amlodipine	24 h	24 h	–	–	1,4-dihydropyridine ring	
Moxonidine	(0.5 M) HCl 80 min	2 M NaOH at 70 °C for 15 min	7 days at 80 °C	H <sub>2</sub> O <sub>2</sub> 5 h	Impurity-A and impurity-D	[148]
Rivaroxaban	(60 °C) 6–24 h	(60 °C) 1–6 h	2–5 h (254 nm)	–	Morpholinone moiety, oxazolidinone ring, chlorothiophene carbonyl.	[149]
Ciprofloxacin, Moxifloxacin, Norfloxacin, Ofloxacin	–	–	105–113 day	–	CP-1 - CP-6 and OP-1 - OP-6, NP-1 - NP-5 MP-1 and MP-10	[150]
Hexazinone	–	–	Degussa P25 and the nano-TiO <sub>2</sub> . 40 min complete degradation	–	Methylamine, triazinone, urea, carbon dioxide.	[151]
Ciprofloxacin, Difloxacin, Lomefloxacin, Norfloxacin, and Ofloxacin	–	–	–	KMnO <sub>4</sub> in acidic pH (3–6)	N-1 and N-4 piperazine atoms and 3- monohydroxylated products 3-hydroxy-5-oxo derivatives	[152]
Ampicillin	(60 °C) 4 h	(60 °C) for 4 h	–	–	2-(acetamidomethyl)-5,5- dimethylthiazolidine- 4-carboxylate 1, 1-dioxide	[153]



**Table 5** Impurity profile of drug using UHPLC-MS/MS

Sr. no	Drug(s)	Detection wavelength	Name of impurity/number of impurities detected	References
1	Finasteride	210 nm	4	[154]
2	Famotidine	265 nm	Imines	[155]
3	Desloratadine	280 nm	5	[156]
4	Sitagliptin, vildagliptin, and metformin	210 nm	2	[157]
5	Heroin	–	Acidic and neutral impurity	[158]
6	Dabigatran Etxilate	220 nm	3	[159]
7	Zolmitriptan	–	4	[160]
8	Smokeless powder	210 nm	Nitroglycerin (NG), 2,4-dinitrotoluene (2,4-DNT), and 2,6-dinitrotoluene (2,6-DNT)	[161]
9	Ritonavir	240 nm	13	[162]
10	25-OCH <sub>3</sub> -PPD	495 nm	3	[163]

reducing the retention of non-basic analytes due to the low ligand density and ionic repulsion [10].

#### Peptide separation technology

The separation or isolation of different peptides, the peptide-based peptide separation technology columns, was utilized for analysis of peptides. Developed peptide separation technology (PST) columns are C<sub>18</sub> BEH Technology, in PST column particles sizes in the variety of 1.7 μm to 10 μm and the column dimension ranges from 75 μm to 30 mm internal diameter and column length from 50 to 250 mm. The PST columns demonstrate sharp-edged symmetrical peaks [7].

#### UHPLC, superior technique than HPLC

The UHPLC system is superior to HPLC system because the UHPLC system operates at high pressure up to 1000 bar or more than that, but a conventional HPLC system, compass a pressure up to 400 bars and it suffers problems like mobile phase swallowing and increases the time of analysis. But in UHPLC system, less solvent consumption and less time for analysis are required [13]. This could unlikely outstanding use of slighter particles less than 2.0 mm and also at tolerable flow up to 5 mL/min. The use of the lesser particle size shows better resolution of peaks, perform faster analysis, sharper, and higher peaks [14].

#### UHPLC-MS/MS detection

To detect and find out the amount of analyte, the mass spectrometry (MS) was widely used because of its selectivity (Watson and Sparkman 2007; Chiu and Muddiman 2008). UHPLC-MS/MS method detection is considered as a method of choice for estimation of drugs in bulk and pharmaceutical formulations and their metabolites

in biological fluids. In literature, there are several methods reported for analysis of drugs using UHPLC-MS/MS. In all ionization techniques, electrospray ionization (ESI) and atmospheric chemical ionization (APCI) ion source were extensively used [15]. Among all kinds of mass analyzers *viz* quadrupole, ion trap, and time of flight, the triple quadrupole was the most preferred one. The several mobile phases were used in UHPLC-MS/MS that includes many solvents such as water, ACN or MeOH, acetic acid, 0.1% formic acid, ammonium hydroxide/ammonia solution, and ammonium acetate (10 mM).

Mass spectrophotometer prominently consists of an ion source to construct gas phase ions and mass analyzers to separate these ions according to their mass to charge ratio; further, a detector to count the ions for every *m/z* ratios. The PC will change over the information from the analyzer and detector to a mass spectrum [16, 17]. The choice of detector is based on the necessary detection sensitivity and it is additionally controlled by other clear necessity; for example, the thermal, chemical, and required stability and the amount of space available. Ideal properties for detectors are high intensification, fast time analysis, low noise, high collection efficiency, minimal effort, narrow distribution of responses, and long life [18]. The MS-MS detectors are more advantageous over the other detectors with respective to the higher sensitivity to detect very small quantity, high selectivity to notify molecules apart in a mixture, and high time resolution and long lifetime. There are different detectors which are used in mass spectrophotometer such as electron multiplier, Faraday cup electrode, and photomultiplier. Electron multiplier is based on that the ion or electrons strikes on the first dynode, resulting in the emission of several electrons. These secondary electrons are then attracted to the second dynode, where

**Table 6** Separation of isomers studies using UHPLC

Drugs/compounds	Column	Mobile phase system	Mode of analysis	Detector	Extraction method	Ref
Flavanone, Hesperetin, and Naringenin	Chiralpak AD-3R	Methanol	Isocratic	DAD	SPE	[165]
Flavanone, Naringenin, and Hesperetin	Chiralpak AD-3R	Methanol	Isocratic	Triple-quadrupole Mass spectrometer	SPE	[166]
Phenylethylamines and cathinones	BEH phenylCSH Fluro-phenyl	80 mM phosphate buffer (A) and acetonitrile or methanol (B)	Isocratic/gradient	PDA	-	[167]
Besifoxacin	Chiralpak IE-3	Dichloromethane: isopropyl alcohol: trifluoroacetic acid: diethylamine (90:10:0.3:0.05 v/v/v/v)	Isocratic	UV-Visible detector	-	[168]
Warfarin	ZORBAX RRHD Eclipse C18, SB-CN, SB-C8 and SB-Phenyl Hypersil Gold AQ and Gold PFP	0.1% Phosphoric acid in water (A) 10 mM ammonium acetate (pH 6.5) (B) and 5mM ammonium phosphate dibasic in water (pH 8.0) (C). 95:5 to 5:95 % A, B or C.	Gradient	PDA	---	[169]
Debenzazepine and Carbamazepine analogs,	Hypersil Gold PFP	2 mM Ammonium formate (A) and acetonitrile (B) 0-0.2 min, 80% (A); 20% (B); 0.2-0.8 min, 5% (A) :95% (B); and back to 80% (A); 20% (B) in 0.5 min	Gradient	PDA	-	[170]
Xanthine metabolites	SB-Phenyl	0.1 H <sub>3</sub> PO <sub>4</sub> in water: acetonitrile	Gradient	PDA	-	
Steroid analogs	Eclipse Plus C <sub>18</sub>	2 mM Ammonium formate in water (pH 3.5) and acetonitrile (65:35 to 5:95 in 2.0 min)	Gradient	PDA	-	
Nicotine analogs	SB-CN	5 mM Ammonium phosphate dibasic in water: acetonitrile	Gradient	PDA	-	
Glucuronide analogs	SB-CN	150 mM sodium perchlorate in 0.02 % perchloric acid: acetonitrile	Gradient	PDA	-	
Rasagiline	Chiralpak AGP	Ammonium acetate: isopropyl alcohol (90:10, v/v)	Isocratic	PDA	--	[171]
Trantinterol	BEH C <sub>18</sub>	3 mM ammonium acetate (A) and acetonitrile (B). 0-4 min, 80-20 % (A); maintained at 70% (A) for 2.5 min	Gradient	Triple-quadrupole mass spectrometer	LLE	[172]
SCH 503034	BEH C <sub>18</sub>	A1-Methanol: water (5:95) with 40 mM PFPA and A2-methanol: water (95:5) with 40 mM PFPA. B1- Methanol: water (5/95) with 4 mM of ammonium acetate and B2-methanol: water (95/5) with 4 mM of ammonium acetate	Gradient	Triple-quadrupole mass spectrometer	PP	[173]
Ranolazine	BEH RP <sub>18</sub>	(A) Acetonitrile: sodium dihydrogen phosphate (pH 7.3; 0.01 M)-Triethylamine (10:90:0.1, v/v/v). (B)- Acetonitrile-mobile phase-A (55:45, v/v) 0.01 min, 17 % (B); 1.5 min, 17%(B); 3.5 min, 45%(B); 5.5 min, 60% (B); 8 min, 65 % (B); 12 min, 70% (B); 13 min, 95% (B); 15 min, 95 % (B); 15.5 min, 17% (B); and 18 min, 17%(B)	Gradient	PDA	-	[174]
Rivastigmine	BEH Phenyl	(A) Acetonitrile: disodium hydrogen orthophosphate (pH 7.5; 0.01 M)-Triethylamine (10:90:0.1 v/v/v). (B) Acetonitrile: water (80:20 v/v) 0.01 min, 5.0 % (B); 5.0 min, 35.0 % (B); 8.0 min, 60.0% (B); 8.1 min, 5.0 % (B); 10.0 min, 5.0%(B).	Gradient	PDA	-	[175]

**Table 6** Separation of isomers studies using UHPLC (Continued)

Drugs/compounds	Column	Mobile phase system	Mode of analysis	Detector	Extraction method	Ref
Benzoic acid	Thermo Synchronis C <sub>18</sub>	0.1% formic acid in water (A) and methanol (B) 0–1.5 min, 15–46 % (B); 1.5–1.9 min, 46% (B); 1.9–4.5 min, 15% (B)	Gradient	Triple-quadrupole mass spectrometer	LLE	[176]
Phenylmercapturic acid and Benzyl mercapturic acid	BEH C <sub>18</sub>	0.1% acetic acid in water (A) and methanol (B). 0–1.0 min, 80% (A); 1.01–6.00 min, 80–70% (A); 6.00 min, 70–80 % (A)	Gradient	Triple-quadrupole mass spectrometer	LLE	[177]
Abacavir	BEH C <sub>18</sub>	Water: Acetonitrile (90:10 v/v)	Isocratic	UV-Vis and PDA	–	[178]
Stimulants agents	BEH C <sub>18</sub>	Water (A) and acetonitrile with 0.01% formic acid (B) (50:50 v/v) Four types of gradient program were used: 1. Gradient A/B: 0 min 1% B; increase to 3% B in 4.2 min, increase to 90% B in 3.8 min, 90% B for 0.5 min, decrease to 1% B in 0.1 min, and equilibration for 1.4 min. 2. Gradient C: 1% B for 1 min, increase to 15% B in 7 min, increase to 90% B in 1 min, 90% B for 1 min, decrease to 1% B in 0.1 min, and equilibration for 1.9 min. 3. Gradient D: 5% B for 1.2 min, increase to 90% B in 16.8 min, 90% B for 0.7 min, decrease to 5% B in 0.1 min, and equilibration for 1.2 min. 4. Gradient E/F: 1% B for 1.2 min, increase to 90% B in 6.8 min, 90% B for 0.7 min, decrease to 1% B in 0.1 min, and equilibration for 1.2 min	Gradient	Triple-quadrupole mass spectrometer	–	[179]
Vigabatrin	Eclipse Plus C18 column	A-10 mM ammonium formate (pH 3.0 adjusted with formic acid and methanol (B), 0–1 min, 10–10% (B); 1–8 min, 10–16% (B), 8–9 min, 16–100% (B), 9–11 min, 100% (B)	Gradient	Triple-quadrupole mass spectrometer	LLE	[180]
Zoxamide	Lux Amylose-2	Acetonitrile: water (70:30 v/v)	Isocratic	Triple-quadrupole mass spectrometer	QuEChERS	[181]

every electron creates a few more electrons, and likewise the electrons are produced. The electron multiplier offers several advantages over the other detectors like low noise, high sensitivity, and typical gain of 106 and total life of electron multiplier 1–2 years. Faraday cup electrode is also known as cylinder electrode. The basic principle behind cylinder electrode is that the incident ion strikes the dynode surface which emits electrons and induces a current which is amplified and recorded. The dynode electrons which are used in faraday cup are made up of secondary emitting material like CsSb, GaP, or BeO. The cylinder electrode is very robust and it is generally used to isotope analysis and isotope-ratio mass spectrometry (IRMS). The photomultiplier are currently presumably the most widely recognized in mass spectrophotometer and it's also known as scintillation counter. The basic principle of scintillation counter or photomultiplier is that the ions at first strike a dynode which outcome in electron emission. These electrons then strike a phosphorous screen which thusly discharges a burst of photons. The photons then pass into the multiplier where amplification occurs in a cascade fashion. The main advantage of using photons is that the multiplier can be kept sealed in a vacuum preventing contamination and greatly extending the lifetime of the detector [16, 19].

#### **Present perspective on UHPLC-MS/MS methods**

The magnitude of the topic on a UHPLC-MS/MS can be learnt from the review articles published on the topics; to our knowledge, more than 11 [20–30] review articles on various facets of UHPLC-MS/MS have been published by different authors. The brief summary of review articles published in different journals on the topic “UHPLC-MS/MS” is depicted in Table 2.

#### **Application of UHPLC-MS/MS**

##### ***Bio-analytical method/metabolite studies***

The sensitivity and selectivity of UHPLC at low detection levels produces precise dependable information that can be utilized for a wide range of purposes, including pharmacokinetics study, toxicity, and bioequivalence studies as the quantification of a medicinally active agent in biological samples is a crucial part of the development program of a bioanalytical method. The sample preparation techniques gained more significance in bioanalytical methods. UHPLC-MS has vital importance in metabolomics and proteomics [31]. A variety of sample preparation techniques have been applied in a bioanalytical method such as protein precipitation, liquid-liquid extraction, and solid-phase extraction [32]. Among these, the most widely used are protein precipitation and liquid-liquid extraction followed by solid-phase extraction which are used in

different category such as antibiotic [33–40], anticancer [36, 41–45, 46–62], antiviral [63–67], antifungal [68–72], cardiovascular [73–86], analgesic [87–93], steroid [94, 95], NSAID [96–106], diabetes [107–110], antidepressant [111–114], antiepileptic [115–119], CNS [120, 121], proton pump inhibitor [122], anti-tubercular [123], vitamin [124, 125], immunosuppressant [126–128], anticoagulant [129, 130], expectorant [102], and multicomponent analysis of drug [131–145]. The simplest is protein precipitation by acetonitrile and centrifugation prior to analysis. The liquid-liquid extraction by ter-butyl methyl ether, chloroform, ethyl acetate, ethyl ether, etc., and solid-phase extraction techniques, disks, or cartridges are used to extract the sample by using methanol, formic acid, water, etc [146]. Bioanalytical studies using UHPLC-MS/MS is summarized in Table 3 and Fig. 2.

##### ***Stability-indicating method using UHPLC-MS/MS***

It is performed at the preliminary stage in the process of drug development [7]. The research conditions force the drug compounds to degrade under tremendous conditions such as acid and base hydrolysis, peroxide oxidation, photo-oxidation, and thermal stability to identify the resultant degradation products. Stability indicating method using UHPLC-MS/MS is included in Table 4 [147–153]

##### ***Impurity profile of drug using UHPLC/MS/MS***

For the medicine development and formulation process, profiling, detection, and evaluation of drug substances and their contamination in crude materials and finishing yield testing is an essential part which is summarized in Table 5 [3] [154–163].

##### ***UHPLC in separation of isomers***

Chirality may influence biological activity; so, there is a necessity for the enantioseparation or diastereoseparation of these compounds. Therefore, sequentially to further examine the mechanisms of action of the enantiomers or diastereomers, it is necessary to understand which compounds are present at the site of action. Therefore, there is an increased demand for sensitive analytical methods to quantify and evaluate the chirality of metabolites present in biological fluids [96]. UHPLC is an incredible tool for the present rehearsing chromatographer, as it can essentially expand the output of a chromatographic separation. What is more, the more extensive scope of usable stream rates makes rapid partitions conceivable. Up till now, various UHPLC columns were employed for separation of structural analogs or for separation of isomers. These UHPLC columns are flexible column with excellent chemical durability and are suitable for quick analysis of samples containing hydrophobic compounds that are strongly retained

in columns or samples containing compounds with large differences in hydrophobicity. In addition, its high bonding density consent for excellent separation of compounds with minute structural differences. Furthermore, these UHPLC columns are used for a wide range of application areas such as measuring the optical purity and purification of chiral materials [164]. The separations of isomers using UHPLC are summarized in Table 6.

### Future direction

In the last decade, literature survey shows that several drugs and metabolites were separated, from their impurities and degradation product by implementing UHPLC-MS/MS technique. For separation or extraction of analytes or drugs and metabolites from the biological fluid with the help of organic solvents, but there are many problems associated with those organic solvents such as toxic nature, injurious to the environment, and dangerous. To overcome these problems, the green solvents are the choice for extraction of analytes. In the future, the more prominent result is also obtained with the help of green solvents because advantage green solvents are their high viscosity, high thermal stability, and low vapor pressure. They are also highly reusable and therefore considered efficient compared to organic solvents. In this survey, the column generally used is BEH C18. The green solvents with the help of increases; high-throughput, sensitivity, and resolution of drugs analysis and identify structures of compounds. Nonetheless, steady up gradation of UHPLC-MS/MS techniques besides data-handling routines are still obligatory for data preprocessing, statistical analysis, biomarker recognition impurity, and degrades products.

### Discussion and conclusion

The presented review article gives a perspective on UHPLC-MS/MS in drug substance and medicine product. The current review article collects a simplified, fast, and selective UHPLC-MS/MS methods which were developed for the determination of drug in bulk and also in plasma. UHPLC-MS/MS technique holds out a very promising system for isolation, characterization, and identification of degradation products and impurities. This knowledge can create abundant information about drugs and guidance for its storage, increasing tools for quality control, and safer treatments.

### Abbreviations

BEH: Ethylene-bridged hybrid; CSH: Charged surface hybrid; HETP: Height equivalent of theoretical plate; HILIC: Hydrophilic interaction chromatography; HPLC: High-performance liquid chromatography; HSS: High strength silica; IRMS: Isotope-ratio mass spectrometry; LC: Liquid chromatography; LLE: Liquid-liquid extraction; MPa: Megapascal; MS: Mass spectrometry; NSAID: Nonsteroidal anti-inflammatory drug; PP: Protein precipitation; Psi: Per square inch; PST: Peptide separation technology; QuEChERS: Quick, easy, cheap, effective, rugged, and safe; RP: Reverse phase;

SB: Selectivity for bases; SPE: Solid-phase extraction; UHPLC: Ultra high-performance liquid chromatography; UPLC-MS/MS: Ultra performance liquid chromatography-mass spectrometry

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