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# Design of experiment-driven stability-indicating RP-HPLC method for the determination of tofacitinib in nanoparticles and skin matrix

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# **Abstract**

**Background:** Tofacitinib—an oral JAK inhibitor—has been recently approved by US FDA to treat moderate to severe RA. The delivery of tofacitinib to specific inflammation site at joint via topical route using nanoformulations helps in managing the potential adverse effects. The objective is to develop and validate a simple, specific, and sensitive stability-indicating HPLC method for quantification of tofacitinib in topical nanoformulations and different matrices (adhesive tape, and skin layers, i.e., stratum corneum, viable epidermis, and dermis). The major objective was to avoid use of instruments like LC–MS/MS and to ensure a widespread application of the method.

**Result:** A  $3^2$  factorial 'design of experiments' was applied to optimize process variables, to understand the effect of variables on peak properties. The calibration curve showed regression coefficient ( $R^2$ ) 0.9999 and linearity in the concentration range of 50 to 15,000 ng/mL, which is suitable for the analysis of conventional dosage forms and nanoformulations. Method validation was performed as per ICH guideline Q2 (R1). The accuracy by recovery studies ranged between 98.09 and 100.82%. The % relative standard deviations in intraday and interday precisions were in the range of 1.16–1.72 and 1.22–1.80%, respectively. Forced degradation studies indicated the specificity of method and showed stability-indicating potential for tofacitinib peak.

**Conclusion:** The validated method provides a quantification method of tofacitinib in the presence of formulation excipients, dissolution media, and skin tissues in detail. In addition, the method was successfully utilized for determination of various dermatokinetics profile of tofacitinib.

Keywords: Dermatokinetic study, Factorial design, HPLC, Nanoparticles, Tofacitinib

# **Background**

Rheumatoid arthritis (RA) is a chronic, systemic inflammatory-mediated and pathologically complex autoimmune disease that can lead to permanent joint damage and deformity [1]. The United States Food and Drug Administration (USFDA) approved tofacitinib, the first

Janus kinase (JAK) inhibitor, in November 2012 for the treatment of moderate to severe RA. Tofacitinib is a selective small-molecule inhibitor of JAK3 and JAK1, and tyrosine kinase (TYK2) to a lesser extent [2]. Importantly, tofacitinib can also modulate T-cell activation, synovitis, and structural joint damage [3]. Tofacitinib is taken twice a day orally in doses of 5 mg [4]. Oral administration of tofacitinib in high doses (5 mg and 10 mg per day) suppresses the immune system and can make the subject susceptible to infections. Thus, there is a need to explore other routes of administration for effective delivery of tofacitinib. Recently, topical application of 2% tofacitinib

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ointment has shown good therapeutic activity for treating psoriasis condition in clinical trials [5]. The topical administration of tofacitinib can overcome the limitations of oral therapy, such as pre-systemic metabolism, gastrointestinal problems, dose escalation, and nontissue distribution and systemic side effects (decrease in neutrophils count). Moreover, compared to the oral route, the topical and transdermal routes require a low dose of tofacitinib, attenuating systemic side effects.

In the case of topical and transdermal delivery, the outermost layer on the skin (stratum corneum) acts as a barrier [6]. The stratum corneum consists of corneocytes in the intercellular matrix form a barrier for the permeation of drugs through the skin. Therefore, lipid-based nanocarriers, viz. solid lipid nanoparticles (SLNs), have been designed to enhance the permeation of drug through the skin [7]. Further, the formulated lipid nanocarriers need to be evaluated for entrapment efficiency, drug release, and permeation through skin layers. Most importantly, to evaluate the topical drug delivery, there is a need to assess the permeation of tofacitinib through different layers of the skin (dermatokinetics). Thus, to understand the dermatokinetics, it is essential to develop a method for determination of tofacitinib in the skin tissues.

A thorough literature survey revealed that there are few tandem mass spectrometry (LC-MS/MS) [8–14], and matrix-assisted laser desorption ionization mass spectrometry imaging (MALDI-MS) [15] methods are available for the estimation of tofacitinib alone and in combination with other analytes. These mass spectrometry methods are sensitive, but the routine estimation of in vitro release samples, and stability samples would be difficult and expensive when the number of samples is more. Additionally, these methods involved very complicated procedures and required additional sample treatment.

The reported HPLC methods for tofacitinib determination showed lower limit of quantification in microgram level only and not fully validated as per ICH guidelines with respect to stability studies and robustness [8, 16-23]. The reported spectrometric methods are economical and adequate for the estimation of tofacitinib in pure form and in its dosage forms [24, 25]. However, these spectrometric methods were not explored for nanogram sensitivity, purity of target peak, and stability. Recently, a fully automated in situ ultraviolet fiber optic system with 10-mm-arch probes for the estimation of TF has been reported [26]. However, this is quite expensive, require accurate installation, a trained operator, and it might be difficult to design acceptable measurement systems for routine analysis. Further, the above reported methods were not investigated and validated in the presence of skin tissue or for complex nanoformulations.

Therefore, in the current work, we developed and validated a rapid, accurate, and precise stability-indicating reverse-phase HPLC method for the estimation of tofacitinib up to 50 nanogram levels in nanocarrier formulation and skin tissues. Further, the design of experiment (DoE) was applied to study the factor interaction and their impact on chromatographic properties [27]. The developed method is suitable for industry and academia for quantification of tofacitinib in nanoformulations, with ease and economic way. It is employed for evaluation of entrapment efficiency, stability, and in vitro release profile of tofacitinib-loaded SLNs. The % recovery of tofacitinib in the skin matrix was calculated, and the method was successfully applied for studying the dermatokinetics of tofacitinib on topical application.

## Methods

# Solvents and chemicals

Methanol and acetonitrile (HPLC grade) were acquired from Merck Limited, Mumbai (India). Ammonium acetate, potassium phosphate monobasic, orthophosphoric acid, and tetra-hydrofuran were acquired from Merck. Milli-Q water was attained in-house from a Milli-Q water purification system, Millipore (USA).

# Instrumentation and chromatographic conditions

The chromatographic experiments were conducted on the HPLC system (Shimadzu, Kyoto, Japan) comprised of model LC-10AT, a binary pump, SIL-HTA autosampler (Shimadzu, Kyoto, Japan), column oven (CTO-10AS) compartment, and SPD-M20A photodiode array (PDA) detector. Chromatographic separation was carried out at  $30 \pm 0.5$  °C using LiChrospher<sup>R</sup> 100 RP-18 analytical column (Hibar<sup>R</sup> 250-4.6; 5 µm; Merck<sup>R</sup>). In the first instance, preliminary trials were conducted to acquire knowledge about the method performance and identification of various critical, independent parameters, and its effect on dependent variables. Systematic method development strategies were applied to identify the independent parameters with less number of trials. Primarily, acetonitrile and methanol were tried with various % ratios of ammonium acetate and phosphate buffers (pH 5, 10 mM) to obtain desired peak symmetry. A 3<sup>2</sup> full factorial experiment design was applied for the optimization of the mobile phase composition. The detection was carried out with a PDA detector with 30 µL injection volume at wavelength 285 nm. The data acquisition and HPLC system were controlled by the LC solution software version 1.24 SP1.

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# **Methods**

# Preparation of stock, calibration, and quality control standards

A stock solution of to facitinib (1 mg/mL) was prepared by dissolving an accurately weighed amount of to facitinib in methanol. The working standard solution (100  $\mu g/mL$ ) and calibration standards 50–15,000 ng/mL were prepared through serial dilution with methanol. The quality control (QC) standards were prepared from the standard stock at three concentration levels: low QC (250 ng/mL), medium QC (8000 ng/mL), high QC (12,000 ng/mL), and the lower limit of quantification (LLOQ) (50 ng/mL).

# DoE methodology and optimization of analytical method

Using the above-executed trials, the organic phase and

$$\label{eq:lodger} \text{LOD and LOQ} = \frac{\text{Concentration of standard}}{\frac{S}{N} \, \text{value of standard}} * \text{Desired} \frac{S}{N} \, \text{value}$$

pH of phosphate buffer demonstrated a high impact on retention time and tailing factor. Therefore, to investigate the effect of organic phase composition and phosphate buffer pH on drug retention time and tailing factor, DoE methodology was applied [28]. A  $3^2$  factorial design consisting of 2 factors at 3 levels was considered for an experimental plan with Design-Expert 8.0 Stat-Ease Inc. Minneapolis, USA. The two independent variables % organic phase  $(X_1)$ , and pH of phosphate buffer  $(X_2)$  with 3 levels (-1 (3.5), 0 (4.5), +1 (5.5)) were confiscated as the actual value. The retention time and tailing factor were considered as dependent variables as responses  $Y_1$  and  $Y_2$ , respectively.

# Validation of the developed method

The validation of the analytical method was performed for system suitability, linearity, range, detection limit, quantification limit, specificity, accuracy, precision, carryover effect, and robustness according to the ICH Q2 (R1) guideline (2005).

# System suitability

System suitability test was preferred for chromatographic methods to ensure that the system is efficient to give reproducible results. The performance of the system was evaluated by injecting six replicates of 10  $\mu g/$  mL concentration with optimized chromatographic conditions.

# Linearity, limit of detection and quantification limit

Linearity was determined with a concentration range between 50 and 15,000 ng/mL with six calibration standards. The obtained data were fitted into linear regression analysis, and the calibration curve was plotted by the analyte peak area on the x-axis against the concentration of analyte on the y-axis. The detection limit (or) limit of detection (LOD) and quantification limit (or) limit of quantification (LOQ) were decided based on the signal-to-noise (S/N) ratio. Initially, in the system suitability test, the signal-to-noise ratio was obtained based on the detector response. The preferred S/N ratios were 3:1 and 10:1 for LOD and LOQ, respectively. LOD and LOQ were calculated based on the below mentioned formula [29]. From these, LLOQ has been determined and considered as the lowest standard of the calibration curve.

# **Accuracy and precision**

The measurement of the observed value's proximity to a given value is known as accuracy. Precision, on the other hand, relates to the closeness of measurement values to one another. The accuracy and precision of the quality control samples LQC, MQC, and HQC, as well as LLOQ, were determined in six replicates on intra- and interdays. Accuracy was denoted as % bias and the precision as % relative standard deviation (RSD). The acceptance requirement for precision and accuracy of quality control samples according to regulatory criteria was  $\leq \pm 2\%$  and  $\leq \pm 10\%$  RSD, respectively [30].

# Carryover effect

The carryover was assessed by analyzing successive samples (10  $\mu$ g/mL, 12  $\mu$ g/mL, and 15  $\mu$ g/mL) of the linearity curve followed by a blank. Carryover acceptance criteria should not be higher than 20% of LLOQ.

# Robustness

Robustness can be defined as the reproducibility potential of the developed method in the same laboratory conditions with slight modification in chromatographic conditions and different HPLC systems with specified conditions. Initially, the robustness of the developed analytical method was performed by changing the column oven temperature, to  $\pm 5$  °C, and mobile phase pH, to  $\pm 0.5$ . Further, the optimized method was tested on another laboratory condition with the Shimadzu system (model number: LC 2010CHT).

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# Specificity

The specificity of the developed analytical method was studied in the presence of formulation excipients (lipids). A known concentration of tofacitinib within the linearity range was spiked into the lipids and analyzed using the developed HPLC method. The interference of lipids with the retention time of analyte and peak purity was observed.

# Stability-indicating property of the developed method

The stability-indicating property of the developed analytical method was studied by exposing the tofacitinib solution to stress conditions as per ICH Q1A (R2) guidelines. The stress studies of tofacitinib solution were conducted under acidic hydrolysis, base hydrolysis, oxidation, and thermolytic conditions [31]. The acid hydrolysis and base hydrolysis were carried out by preparing the tofacitinib solution (200 µg/mL) using 0.5 M Hydrochloric acid and 0.5 M sodium hydroxide and kept for reaction on a water bath at 60 °C for 6 h. Similarly, for oxidative degradation, tofacitinib solution was prepared using 2% hydrogen peroxide and heated at 60 °C under reflux condition for 3 h. For thermal degradation, the tofacitinib solution was heated at 80 °C under reflux condition for 6 h. After subjecting tofacitinib solution (200 μg/mL) to the abovesaid stress conditions, a concentration of 10 µg/mL was prepared using methanol. The samples of acid and base were neutralized before dilution with methanol to protect the column. All the samples were filtered through a 0.2-µm filter before injecting it into HPLC analysis [32, 33]. The chromatogram of different stress conditions was recorded and compared with the normal condition. The retention time of different degradant peaks and drug peaks was identified and calculated the % degradation.

# Applicability of the developed method for skin studies of tofacitinib and dermatokinetic assessment

The validated method for quantification of tofacitinib in the presence of skin tissue matrix was used to evaluate tofacitinib penetration through skin. Initially, skin tissue was homogenized using an Ultra-Turrax type homogenizer. Furthermore, the homogenized tissue was spiked with a predetermined concentration of tofacitinib solution (500  $\mu$ g/mL) and centrifuged. The supernatant (1 mL) was collected, and from this a concentration of 10  $\mu$ g/mL was prepared. The samples were filtered through the 0.22- $\mu$  membrane filter before analysis and observed for peak specificity, and percent recovery was calculated [34]. Inhouse tofacitinib cream was prepared using stearic acid as base (0.5 mg/g) and applied topically (350 mg). After topical application, skin samples were collected for each time point at 2, 4, 6, 8, 12, 24 h. The collected skin samples were

washed with phosphate buffer and gently wiped with the cotton. Tape stripping analysis was performed to separate the epidermis and dermis layers. The collected tapes and skin were soaked for 6 h in methanol. After 6 h, the samples were filtered through 0.25- $\mu$  filter and then analyzed at – 20 °C until analyzed as per validated method. To facitinib concentration in epidermis and dermis–time profiles were analyzed by non-compartmental model approach to determine  $t_{1/2}$ , half-life;  $C_0$ , to facitinib concentration in epidermis and dermis at t=0; AUC $_{0-v}$ , area under the curve from zero to the last measurable point; AUC $_{0-\infty}$ , area under curve from time 0 extrapolated to infinity.

# Applicability in characterization of nanoformulation

The validated analytical method was solicited for its use in the characterization of SLNs. The tofacitinib-loaded SLNs were prepared by the hot emulsification technique [35, 36]. The SLNs were prepared using Precirol as solid lipid and Poloxamer 407 as a surfactant as per the reported method. The prepared formulation was analyzed for its entrapment efficiency, in vitro drug release (pH 7.4), and stability studies.

# **Entrapment efficiency**

The entrapment efficiency of prepared SLNs formulation was estimated by indirect method. In brief, the formulation was subjected to ultracentrifugation using Remi cooling centrifuge (Mumbai, India). The clear supernatant was diluted with the methanol and analyzed by the validated method.

# % Entrapment Efficiency

 $= 1 - \frac{\text{Free tofacitinib in nanocarrier dispersion}}{\text{Total tofacitinib in nanocarrier dispersion}} * 100$ 

# In vitro drug release studies

The in vitro release of tofacitinib from SLNs dispersion was estimated using the dialysis bag technique. The study was performed using pH 7.4 phosphate buffer with 0.15% sodium lauryl sulfate. A known amount of formulation was transferred into the dialysis bag and maintained at  $32\pm0.5$  °C. The samples were collected at regular intervals 1, 2, 4, 6 h and replaced with the fresh buffer to maintain sink condition. The tofacitinib concentration was assessed using an approved method after the samples were filtered through a 0.22-m filter.

# Stability studies

Pharmaceutical formulation development involves stability at storage conditions to maintain drug activity. Thus, short-term stability of prepared SLNs formulation was carried out at room temperature and controlled condition for

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three months. The formulation was tested for entrapment efficiency after three months of storage period using the validated analytical method.

## Results

# DoE methodology and optimization of analytical method

Preliminary studies were performed for the development of HPLC method. Initially, different combinations of mobile phases were explored using methanol, acetonitrile, and aqueous buffers (potassium phosphate and ammonium acetate - 10 Mm). Peak splitting and broad peaks were observed in the case of acetonitrile, whereas peak symmetry was found acceptable in the case of methanol. Methanol was combined with different proportions of aqueous buffers, and chromatogram properties (peak area, tailing factor, and retention time) were observed. The peak properties were analyzed, and it was observed that methanol in combination with 10 mM phosphate buffer showed good peak properties. The above-mentioned preliminary studies were performed with the agueous mobile phase in 50:50  $\nu/\nu$ . The injection volumes of 20 µL and 30 µL were screened, and it was found that 30 µL showed good peak area with reduced tailing factor. The above trials revealed that the peak properties were mostly influenced by the methanol percentage and pH of the aqueous phosphate buffer. Further, to optimize these parameters, full factorial designed was applied.

A total of 11 experimental runs were performed based on 3<sup>2</sup> factorial design, and obtained results were analyzed for the retention time and tailing factor. The results of the responses are specified in Table 1. To evaluate the relation between the dependent and independent variables, the response surface methodology plots were produced

**Table 1** Suggested experiments by design of experiments and responses

| Variable 1<br>Organic phase | Variable 2<br>Buffer pH | Response 1<br>Retention time | Response 2<br>Tailing factor |  |
|-----------------------------|-------------------------|------------------------------|------------------------------|--|
| %                           | -                       | Min                          | -                            |  |
| 40                          | 3.5                     | 7.243                        | 1.681                        |  |
| 50                          | 3.5                     | 4.918                        | 1.514                        |  |
| 60                          | 3.5                     | 3.667                        | 1.325                        |  |
| 40                          | 4.5                     | 9.790                        | 1.652                        |  |
| 50                          | 4.5                     | 5.430                        | 1.473                        |  |
| 60                          | 4.5                     | 3.868                        | 1.352                        |  |
| 40                          | 5.5                     | 11.168                       | 1.435                        |  |
| 50                          | 5.5                     | 5.630                        | 1.365                        |  |
| 60                          | 5.5                     | 3.902                        | 1.334                        |  |
| 50                          | 4.5                     | 5.420                        | 1.472                        |  |
| 50                          | 4.5                     | 5.430                        | 1.455                        |  |

from Design expert software, as reported in Fig. 1. The experimental runs' responses were fitted into linear, second-order, and quadratic models. Quadratic models proved to be the best fit with p < 0.0001. The best fit model was found to be quadratic models (p < 0.0001). The model summary statistics suggested the quadratic model for responses, because of the low prediction error sum of squares (PRESS) value of the quadratic model. Here, the low standard deviation, high adjusted  $R^2$  value, specifies the good correlation of the fitted model with experimental data. The model was analyzed and validated by analysis of variance (ANOVA), and the results are shown in Table 2.

The independent variables exhibited a reasonable impact on retention time. ANOVA analysis (Table 2) showed model F-value of 80.51 (p value < 0.0001), stipulated that the model was significant. The 3D plots, 2D contour plots, and final polynomial equation (1) for actually coded variables showed the relationship between % of the organic phase  $(X_1)$  and pH of phosphate buffer  $(X_2)$ , on the retention time  $(Y_1)$  (Fig. 1). Figure 1a, b shows that the methanol ratio in the mobile phase had shown a notable effect on retention time. The retention time was decreased with an increase in methanol percentage in mobile phase composition. In contrast, increasing the pH of phosphate buffer resulted in increased retention time. The combination of both the variables  $X_1$  and  $X_2$  showed a parabolic effect on the retention time. The obtained polynomial equation (1) has confirmed the same.

The polynomial equation (1) showed that the % of the organic phase  $(X_1)$  showed a negative effect while pH of phosphate buffer  $(X_2)$  showed a positive effect, but the combination of both the independent variables exhibited a negative effect on retention time  $(Y_1)$ .

Retention time 
$$(Y_1) = +22.67376 - 1.17143X_1 + 7.65975 X_2 - 0.092250X_1X_2 + 0.013072X_1^2 - 0.24834X_2^2$$
 (1)

The independent variables showed a reasonable impact on the tailing factor. ANOVA analysis showed model F-value of 50.03 with p value 0.0003, indicating that the model was significant (Table 2). The 3D plots and final polynomial equation (2) for actual coded variables showed the relationship between % of the organic phase ( $X_1$ ) and pH of phosphate buffer ( $X_2$ ), on the tailing factor ( $Y_2$ ) (Fig. 1a, b).

Figure 1c, d shows that the methanol ratio in the mobile phase had shown a prominent effect on the tailing factor. The obtained polynomial equation (2) has proved the same.

The polynomial equation (2) showed that the % of the organic phase ( $X_1$ ) showed a negative effect, while the

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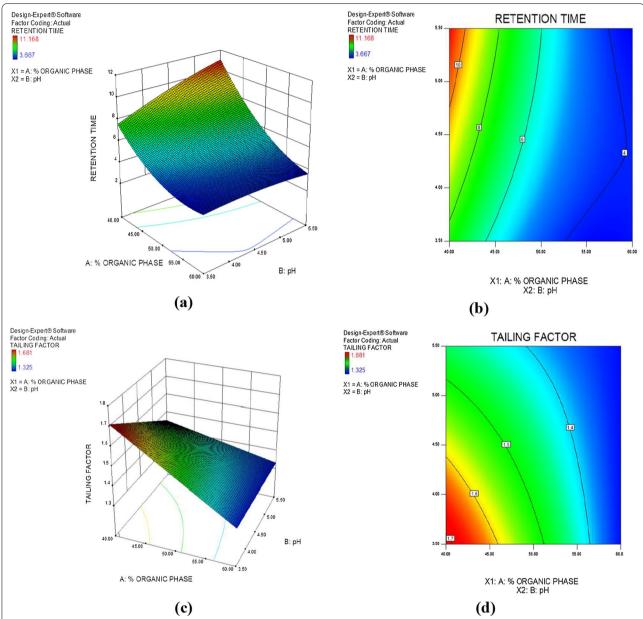


Fig. 1 a, b represent the 3D response curve and 2D contour plots for response 1, i.e., retention time; c, d represent the 3D response curve and 2D contour plots for response 2, i.e., tailing factor

pH of phosphate buffer  $(X_2)$  exhibited a positive effect, but the combination of both the independent variables exhibited a positive effect on the tailing factor  $(Y_2)$ .

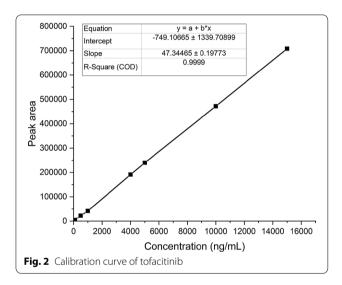
The  $3^2$  factorial design presented 50 solutions for the optimized chromatographic conditions, but the solutions were reduced by setting the goal values. The optimized

Tailing factor 
$$(Y_2) = +3.42085 - 0.060278X_1$$
  
  $+ 8.65351E - 003X_2 + 6.37500E - 003X_1X_2$   
  $+ 1.89737E - 004X_1^2 - 0.043526X_2^2$  (2)

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**Table 2** ANOVA for response surface quadratic model

| Parameter      | Response 1 (Retention time) | Response<br>2 (Tailing<br>factor) |
|----------------|-----------------------------|-----------------------------------|
| Std. Dev.      | 0.38                        | 0.024                             |
| Mean           | 6.04                        | 1.46                              |
| C.V. %         | 6.31                        | 1.63                              |
| R-squared      | 0.9877                      | 0.9804                            |
| Adj R-squared  | 0.9755                      | 0.9608                            |
| Pred R-squared | 0.8712                      | 0.8210                            |
| Adeq precision | 26.389                      | 21.688                            |
| F value        | 80.51                       | 50.03                             |
| p value        | < 0.0001                    | 0.0003                            |



chromatogram conditions were found to be methanol and 10 mM phosphate buffer (pH 3.5) in the ratio of 50:50% v/v with a flow rate of 0.8 mL/min.

# Validation of the developed method System suitability

The system suitability was estimated by six replicate injections of 10 µg/mL concentration of to facitinib with the optimized chromatographic conditions. The tailing factor was found to be 0.997, which indicated the acceptability of peak properties. The % RSD for peak area was within  $\pm 2\%$  indicated the suitability of the system. The standard calibration curve of to facitinib was constructed. The linearity between the concentration of to facitinib and peak area was obtained in the range of 50 ng/mL to 15 µg/mL with an excellent regression coefficient of 0.999. The obtained linear regression coefficient was y=47.34~x-749.10. In the regression equation, 'x' is the concentration of to facitinib and 'y' is the peak area at 285 nm. The calibration curve of tofacitinib is presented in Fig. 2. The LOD and LOQ of the developed analytical method were determined using the signal-to-noise ratio, and values were found to be 16.5 and 49.5 ng/mL, respectively. The LLOQ was found to be 50 ng/mL (n=6). The results proved that the developed method was sensitive enough to detect and quantify tofacitinib in nanogram levels. This method would be beneficial for routine analysis of tofacitinib in nanoformulation where the entrapped drug is in low concentration.

# **Accuracy and precision**

The accuracy and precision were estimated by standard addition method with three concentrations of LQC (250 ng/mL), MQC (8000 ng/mL), and HQC (12,000 ng/mL). The chromatograms representing LQC and HQC are presented in Fig. 3. The method demonstrated acceptable % recovery and reproducibility with % RSD and % bias less than 2%. Intra- and inter-day accuracy and precision data of tofacitinib are depicted in Table 3.

# Carryover effect

The carryover effect was estimated by analyzing successive samples, i.e.,  $10~\mu g/mL$ ,  $12~\mu g/mL$ , and  $15~\mu g/mL$  concentration of tofacitinib followed by blank sample. There was no tofacitinib peak observed in the blank at the retention time of 6.1 min. The result demonstrated that no carryover effect was observed; thus, this method could be utilized for an unremitting run for more number of samples.

The optimized chromatographic conditions were evaluated for robustness with the deliberate change in oven temperature and mobile phase pH. The effect was found to be insignificant with these two variables on analysis of LQC, MQC, and HQC. The RSD (%) was less than 2 and showed good reproducibility of peak properties. The obtained result implied that the developed method was stable against small variations in intrinsic parameters.

# Robustness

The robustness of the analytical method was performed on another Shimadzu system (model number: LC 2010CHT) with the optimized chromatogram conditions. The results demonstrated that there was no change in peak area and tailing factor. The observed changes were insignificant; hence, this method could be easily transferable from one laboratory to another laboratory.

# Specificity

The specificity study was performed with the presence of lipids used for SLNs preparation. There was no peak interference observed at the retention time of the

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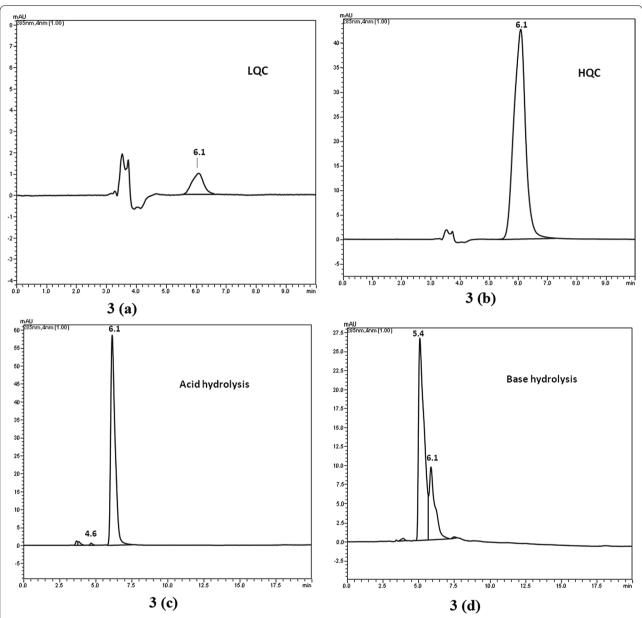
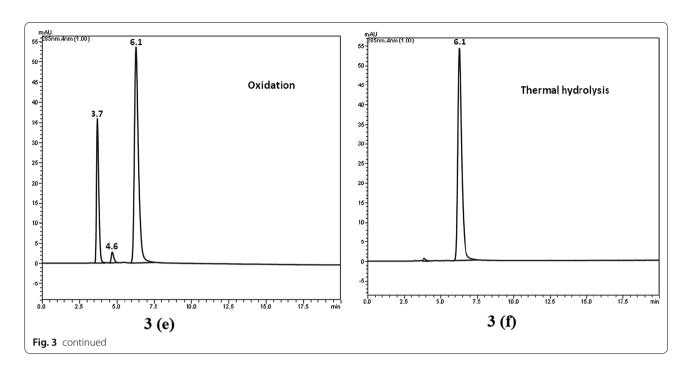


Fig. 3 Representative HPLC chromatograms for a lowest quality control (LQC ~ 250 ng/mL), b highest quality control (HQC ~ 12,000 ng/mL, c acid hydrolysis, d base hydrolysis, e oxidation, f thermal conditions

tofacitinib peak, and the purity of peak was found to be 99.99%. The forced degradation studies such as acid hydrolysis, base hydrolysis, oxidation, and thermal were successfully carried out. The stability results of tofacitinib are shown in Table 4. The chromatogram of acid hydrolysis (Fig. 3c) sample of tofacitinib showed degradation peaks at 3.67, 3.8, and 4.6 min. There was 14.78% degradation of tofacitinib observed under acidic hydrolysis. The chromatogram of base hydrolysis (Fig. 3d) sample of tofacitinib showed peak splitting, and the impurity

peak was prominent and eluted at the retention time of 5.5. This indicates that nearly 75% of to facitinib was degraded. The observed base degradation result was confirmed with the reported study by Youn is et al. They reported that above the pH  $\sim$  9, to facitinib citrate showed the highest degree of degradation [37]. In contrast, under oxidative stress conditions, the chromatogram (Fig. 3e) of to facitinib generated impurity peaks at 3.7 and 4.6 min, and 20.25% degradation of to facitinib was observed. The thermal degradation samples of to facitinib did not show Gorantla et al. Futur J Pharm Sci (2021) 7:180 Page 9 of 12



**Table 3** Intra- and inter-day precision and accuracy results (n = 6)

| Sample | Intraday precision                     |                  | Accuracy            |                  |
|--------|--|------------------|---------------------|------------------|
| QC     | Experimental concentration (μg/<br>mL) | % Recovery ± SD  | % RSD               | % Bias           |
| LQC    | 252.06 ± 4.32                          | 100.81 ± 1.73    | 1.1666±0.019        | 0.818±0.39       |
| MQC    | $7847.82 \pm 192.2$                    | $98.09 \pm 1.52$ | $1.7295 \pm 0.0423$ | $1.908 \pm 0.20$ |
| HQC    | $11,781.5 \pm 213.9$                   | $98.17 \pm 1.25$ | $1.2827 \pm 0.023$  | $1.827 \pm 0.56$ |
|        | Interday precision                     |                  |                     |                  |
| LQC    | 252.06±4.32                            | 100.82 ± 1.44    | 1.6498 ± 0.027      | 0.824±0.39       |
| MQC    | $7915.25 \pm 96.84$                    | $98.94 \pm 1.35$ | $1.2219 \pm 0.014$  | $1.059 \pm 0.20$ |
| HQC    | $11,797.79 \pm 212.75$                 | $98.31 \pm 1.43$ | $1.8017 \pm 0.032$  | $1.685 \pm 0.43$ |

LQC: 250 ng/mL; MQC: 8000 ng/mL; HQC: 12,000 ng/mL

**Table 4** Stability results of tofacitinib in forced degradation study

| Study performed | Stress condition        | % of Tofacitinib degraded | Degradation peaks with retention time (min) |
|-----------------|-------------------------|---------------------------|---|
| Acid hydrolysis | 0.5 M Hydrochloric acid | 14.78 ± 0.04%             | 3.67, 3.8 and 4.6                           |
| Base hydrolysis | 0.5 M sodium hydroxide  | $75 \pm 0.06\%$           | 5.4   |
| Oxidative       | 2% hydrogen peroxide    | $20.25 \pm 0.04\%$        | 3.7 and 4.6                                 |
| Thermal         | Water                   | $1.5 \pm 0.02\%$          | =   |

any impurity peak (Fig. 3f). The sample showed 1.5% degradation of to facitinib.

The results confirmed that the developed analytical method could discriminate the degradation products

from the retention time of tofacitinib. The developed method was found to be selective for tofacitinib and indicated the method stability-indicating property. This can Gorantla et al. Futur J Pharm Sci (2021) 7:180 Page 10 of 12

be applied for stability testing of tofacitinib in-process and finished products.

# Applicability of the developed method for skin studies of tofacitinib and dermatokinetic assessment

The appropriateness of the method was confirmed in dermatokinetics studies on topical administration of tofacitinib. The half-life for tofacitinib was found to be 6 h. The trapezoidal rule was employed for the calculation of AUC <sub>0-last</sub> for tofacitinib. Considering the obtained results, the proposed and validated method could be widely used for their routine analysis and for understanding the in vivo dermatokinetics studies.

# Applicability in characterization of nanoformulation

The applicability of the validated analytical method was appraised by determining the percentage entrapment efficiency, cumulative drug release of tofacitinib from the SLNs, stability, and estimation of tofacitinib from the skin tissues. Tofacitinib-loaded SLNs were analyzed in triplicate for the entrapment efficiency. The entrapment efficiency of tofacitinib was found to be  $96.44 \pm 0.62\%$ . Further, the stability of the prepared nanoparticles was determined by estimating the entrapment efficiency of tofacitinib-loaded SLNs after 3 months. The entrapment efficiency after 3 months was found to be  $76.37 \pm 1.31\%$ , whereas controlled samples showed 99.65% of the initial sample. The validated method was successfully determined the change in the entrapment efficiency of the formulation. Thus, this method can be used to estimate the shelf life of tofacitinib in nanoformulation. The in vitro release is acquiring considerable interest as a surrogate test for product capability. The in vitro drug release samples were collected and filtered through the 0.22-μ membrane, and the concentration of tofacitinib was determined by the developed method (n=3). The cumulative in vitro drug release from nanoparticles was found to be  $34.45 \pm 1.65\%$  at 6 h. The developed method appropriately quantified the tofacitinib in the lipid nanoformulations from the first hour until the end of the release study. The total amount of drug was found to be 98.9%, and the unreleased drug from SLNs was found to be  $64.55 \pm 0.98\%$ . The developed method showed good agreement for the characterization of drug release in dissolution media. Therefore, the developed method can be used for the routine assessment of the effect of formulation factors and quality control. The skin tissue samples were analyzed in six replicates for the determination of % recovery of tofacitinib. The % recovery of tofacitinib from the skin tissue was found to be  $95.61 \pm 0.96$ , with peak purity of  $99.99 \pm 0.54\%$ . A specific peak was observed at 6.1 min without any interference with the skin matrix. The obtained results confirmed that the developed method can be applied for the quantification of tofacitinib during skin permeation and retention studies, dermatokinetics studies.

#### Discussion

The aim of this research work is to improve and validate the RP-HPLC method for estimating tofacitinib in nanoparticles and dermatokinetics studies. The preliminary trials revealed that the peak properties were mostly influenced by the percentage of organic phase and pH of the aqueous phosphate buffer. Using the principle of optimization, we set out to improve the process. The application of the DoE approach allowed for a systematic and easy screening of method variables [38]. Response surface methodology-based 32 factorial design was used to confirm the optimum organic phase composition in mobile phase, buffer pH for the analytical method and to ascertain a mathematical relationship between variables and responses (tailing factor and retention time) [39]. The tailing factor was decreased with an increase in methanol percentage in mobile phase composition, while the pH of phosphate buffer also showed a significant decrease in the tailing factor [40].

The optimized chromatogram conditions exhibited retention time (6.1 min), which is a time-saving RP-HPLC method. The analytical methods involving expensive high-cost equipment complicated sample preparation procedures are difficult to apply in routine laboratories [10, 14, 15]. However, the previously reported spectrometric methods could not confirm the sensitivity in nanogram level, peak purity, and tofacitinib-specific chemical stability. The current validated method has additional advantages over previously reported methods such as short run time, low flow rate, and simple sample preparation procedure and validated as per ICH guidelines with respect to stability-indicating property, robustness, and specificity [20, 25]. Intraday and interday precision at all QC levels of tofacitinib showed that the % RSD of developed methods were in the range of  $1.1666 \pm 0.019\%$  to  $1.7295 \pm 0.0423\%$  and  $1.2219 \pm 0.014\%$ to  $1.8017 \pm 0.032\%$ , respectively. The % bias on intraday and interday was found in the range of  $0.818 \pm 0.39\%$  $1.908 \pm 0.20\%$  and  $0.824 \pm 0.39 - 1.685 \pm 0.43\%$ , respectively. The acceptable values of % RSD and % bias less than 2% indicated that the validated RP-HPLC method is reliable and precise and is in excellent accordance with the regulatory guidelines [41].

The obtained LOD and LOQ values were found to be 16.5 and 49.5 ng/mL indicating that this method has a higher sensitivity than previously reported RP-HPLC methods for tofacitinib [16]. The specificity study confirmed that the developed analytical method could quantify tofacitinib in the presence of formulation excipients.

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The appropriateness of the method was also confirmed in dermatokinetics studies on topical administration of tofacitinib [42]. Finally, the validated method was applied to quantify to facitinib during solid lipid nanoparticle characterization, skin retention studies, and dermatokinetics studies [43]. The method quantified the tofacitinib in skin layers after application of topical formulation. After topical application of (350 mg) conventional cream, the epidermal concentration and dermal concentration of tofacitinib were slowly reached to  $C_{\text{max}}$  of 9.88 µg/cm<sup>2</sup> and 19.95  $\mu$ g/cm<sup>2</sup>, respectively, at 6 h ( $t_{max}$ ). The AUC  $_{0-\infty}$  in epidermal and viable skin layers was found to be 224.78 μg/cm<sup>2</sup> and 345.04 μg/cm<sup>2</sup>, respectively. Based on the findings, the proposed and validated method could be widely utilized for regular analysis and to quantify tofacitinib in in vivo dermatokinetics experiments [44].

## Conclusion

A simple, sensitive, reproducible, robust, cost-effective stability-indicating analytical method has been developed effectively using the Design of experiments for quantification of tofacitinib in pharmaceutical conventional dosage forms and nanoformulations. The optimized chromatographic conditions were fully validated as per ICH Q2 (R1) guidelines and found to be economical for the routine analysis in laboratory conditions compared to the reported LC-MS and HPLC methods. The stabilityindicating studies showed distinct peaks from drug peak during the acid, base, oxidative, and thermal hydrolysis conditions. The validated analytical method evidenced its utility in the estimation of tofacitinib in the pharmaceutical nanoformulations and different skin layers (viable epidermis and dermis). Thus, the developed method can be utilized for industry and academia for tofacitinibloaded nanoformulations characterization and stability evaluation.

# Abbreviations

USFDA: United States Food and Drug Administration; TYK2: Tyrosine kinase; LC–MS/MS: Liquid chromatography with tandem mass spectrometry; RP-HPLC: Reverse-phase high-performance liquid chromatography; SLNs: Solid lipid nanoparticles; DoE: Design of experiment; QC: Quality control; LLOQ: Lower limit of quantification; LOD: Limit of detection; LOQ: Limit of quantification; S/N: Signal to noise; RSD: Relative standard deviation; PRESS: Prediction error sum of squares; ANOVA: Analysis of variance.

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# Plant authentication

Not applicable.

# Authors' contributions

SG is the methodology, investigation, and writing—original draft; RNS is the conceptualization, resources; GS is the conceptualization, investigation,

project supervision, writing—review and editing. All authors read and approved the final manuscript.

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#### **Declarations**

## Ethics approval and consent to participate

The article does not contain any studies with human or animal subjects performed by any of the authors.

#### Consent for publication

Not applicable.

#### Competing interests

No competing interests.

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